

Energy Changes in Chemical reactions for KSS Science - Worksheet (Answer)

- In an exothermic reaction does the temperature go up or down?
The temperature will increase
 - In an endothermic reaction does the temperature go up or down?
The temperature will decrease
 - Name two examples of exothermic reactions.
Respiration & combustion
 - Name two examples of endothermic reactions.
Photosynthesis & thermal decomposition
 - Copy the correct answers.
The words between the arrows of the reactions usually mean to the student that the
is an exothermic reaction. Therefore, these are words between the arrows
of the exothermic reactions. This is an exothermic reaction, correct.
 - Use the table to answer the question.
- | Reaction | Starting temperature /°C | Final temperature /°C |
|----------|--------------------------|-----------------------|
| (a) | 20 | 30 |
| (b) | 20 | 10 |
| (c) | 20 | 20 |
- Which reaction was exothermic? exothermic is exothermic, correct. How can you
confirm?
a. Exothermic temperature increases
b. Exothermic temperature decreases
c. Exothermic temperature increases
 - Which reaction has the biggest energy change?
a. biggest temperature change
 - In an exothermic reaction, is enthalpy change positive or negative?
negative
 - In an endothermic reaction, is enthalpy change positive or negative?
positive
 - When hydrochloric acid reacts with ammonium hydroxide to produce the
temperature goes up
 $\text{HCl} + \text{NH}_4\text{OH} \rightarrow \text{NH}_4\text{Cl} + \text{H}_2\text{O}$
up is - exothermic
Complete the energy profile diagram and state whether the reaction is exothermic
or endothermic, explain your answer.

Energy And Chemical Change Answers

Morris Hein,Susan Arena,Cary Willard



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