

Name: _____

Date: _____

Empirical vs. Molecular Formulas

Compound	Empirical Formula	Molecular Formula
Water	H_2O	H_2O
Hydrogen Peroxide	HO	H_2O_2
Acetylene	CH	C_2H_2
Benzene	CH	C_6H_6
Glucose	CH_2O	$\text{C}_6\text{H}_{12}\text{O}_6$
Acetic Acid	CH_2O	$\text{C}_2\text{H}_4\text{O}_2$

Answer the following question:

1. What is the difference between an empirical formula and a molecular formula?
2. How can you determine the molecular formula from the empirical formula?

Empirical Formulas And Molecular Formual Answers

Pamela Mills,Amina El-Ashmawy



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relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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