

# Empirical and Molecular Formulas Practice

Read the question carefully for the data provided. Answers must be shown for credit.

Find empirical formula:

1. A compound is found to contain 40.94 g of iron and 35.48 g of sulfur. Find the empirical formula.

$$\text{40.94 g Fe} \times \frac{1 \text{ mol Fe}}{55.845 \text{ g Fe}} = 0.7333 \text{ mol Fe}$$

$$\text{35.48 g S} \times \frac{1 \text{ mol S}}{32.06 \text{ g S}} = 1.107 \text{ mol S}$$

FeS

2. A compound is found to contain 20.0 g of carbon, 26.7 g of hydrogen, and 66.7 g of oxygen. Find the empirical formula.

$$\text{20.0 g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 1.66 \text{ mol C}$$

$$\text{26.7 g H} \times \frac{1 \text{ mol H}}{1.008 \text{ g H}} = 26.5 \text{ mol H}$$



$$\text{66.7 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 4.17 \text{ mol O} \left( \frac{4.17}{1.66} = 2.5 \right) \times 2 = 5 \text{ mol O} = 5$$

3. A 10.0 g sample of a compound contains 2.0 g of carbon, 1.0 g of hydrogen, and 7.0 g of oxygen. Find the empirical formula.

$$\text{2.0 g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 0.166 \text{ mol C}$$



$$\text{1.0 g H} \times \frac{1 \text{ mol H}}{1.008 \text{ g H}} = 0.99 \text{ mol H} \left( \frac{0.99}{0.166} = 6 \right) \times 6 = 6$$

$$\text{7.0 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 0.437 \text{ mol O} \left( \frac{0.437}{0.166} = 2.63 \right) \times 2.63 = 7$$

4. A 20.00 g sample of an unknown compound contains 12.00 g carbon, 2.00 g hydrogen, and 6.00 g oxygen. Find the empirical formula.

$$\text{12.00 g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 1.00 \text{ mol C}$$



$$\text{2.00 g H} \times \frac{1 \text{ mol H}}{1.008 \text{ g H}} = 1.98 \text{ mol H}$$

$$\text{6.00 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 0.375 \text{ mol O} \left( \frac{0.375}{0.125} = 3 \right) \times 3 = 3$$

# Empirical Formula Pogil

**S Nieto**



## Empirical Formula Pogil:

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## **Empirical Formula Pogil** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Empirical Formula Pogil**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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