Empirical and Molecular Formula Practice KEY

Empirical Formula

36.5% Na, 25.4% S, and 38.1% O forms this compound.

Na - 36.5/23 = 1.56/0.8 = 2 S - 25.4/32 = 0.8/0.8 = 1

0-38.1/16 = 2.4/0.8 = 3

Na₂SO₃

A compound made of 72.2% Magnesium and 27.8% nitrogen by mass.

Mg - 72.2/24.3 = 3/2 = 1.5(2) = 3

N - 27.8/14 = 2/2 = 1(2) = 2

MigaNa

A chromium compound with 68.4% Cr and 31.6% O.

Cr - 68.4/52 = 1.315/1.315 = 1(2) = 2

O - 31.6/16 = 1.975/1.315 = 1.5(2) = 3

Cr₂O₃

4. A compound with the composition of 24% C, 7% H, 38% F, and 31% P.

C - 24/12 = 2/1 = 2

H - 7/1 = 7/1 = 7

F - 38/19 = 2/1 = 2

P - 31/31 = 1/1 = 1

C₂H₂F₂P

5. 63.5% silver, 8.2% nitrogen, and 28.3% oxygen make up this compound.

Ag - 63.5/107.87 = 0.589/0.585 = 1

N - 8.2/14.01 = 0.585/0.585 = 1

O-28.3/16 = 1.769/0.585 =3

AgNO₁

8.81g C and 91.2g Chlorine forms this compound.

C - 8.81/12 = 0.73/0.73 = 1(2) = 2

CI - 91.2/35 = 2.61/0.73 = 3.5(2) = 7

C₂Cl₂

7. A compound in a soft drink that has 0.31g hydrogen, 3.2g phosphorus, and 6.5g oxygen.

H - 0.31/1 = 0.31/.1 = 3.1

P - 3.2/31 = 0.10/.1 = 1

0 - 6.5/16 = 0.40/.1 = 4

HIPO.

Empirical And Molecular Formula Quiz And Key

Huangqi Zhang

Empirical And Molecular Formula Quiz And Key:

Unveiling the Magic of Words: A Review of "Empirical And Molecular Formula Quiz And Key"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Empirical And Molecular Formula Quiz And Key**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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