

# Elementary Principles of Chemical Processes Solutions Manual 3rd Update Edition Edition Richard M. Felder

## CHAPTER TWO

2.2 (a)  $\frac{1.25 \text{ kg}}{1 \text{ min}} \times \frac{1 \text{ h}}{60 \text{ min}} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ min}}{60 \text{ s}} = \frac{1.25 \times 1000}{60 \times 60} \text{ g/s} = (3.472 \times 10^{-2}) \text{ g/s}$

(b)  $\frac{0.5 \text{ kg} \times 10^{-3} \text{ s}}{1 \text{ min}} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ min}}{60 \text{ s}} = \frac{0.5 \times 1000}{60} \text{ g/s} = (8.33 \times 10^{-3}) \text{ g/s}$

(c)  $\frac{0.04 \text{ kg}^2}{1 \text{ kg}} \times \frac{1 \text{ h}}{60 \text{ min}} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ min}}{60 \text{ s}} = \frac{0.04 \times 1000}{60 \times 60} \text{ g/s} = (1.11 \times 10^{-3}) \text{ g/s}$

2.3 (a)  $\frac{100 \text{ mm}}{1 \text{ s}} \times \frac{1 \text{ m}}{1000 \text{ mm}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ m/s} = (2.78 \times 10^{-5}) \text{ m/s}$

(b)  $\frac{100 \text{ mm}}{1 \text{ s}} \times \frac{1 \text{ m}}{1000 \text{ mm}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ m/s} = (2.78 \times 10^{-5}) \text{ m/s}$

(c)  $\frac{1.27 \times 10^{-2} \text{ kg} \times 1 \text{ min}}{1 \text{ s}} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ min}}{60 \text{ s}} = \frac{1.27 \times 10^{-2} \times 1000}{60} \text{ g/s} = (0.212 \times 10^{-1}) \text{ g/s}$

2.4 Assume that a golf ball (mass = 0.045 kg) is launched from a 2 m × 2 m × 2 m volume. If we assume a 100% efficiency, we can calculate the maximum velocity of the ball.

$m_{\text{ball}} = 0.045 \text{ kg} = 0.045 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 45 \text{ g}$

The maximum velocity is given by the equation:  $v = \sqrt{\frac{2E}{m}} = \sqrt{\frac{2 \times 100 \text{ J}}{0.045 \text{ kg}}} = 66.7 \text{ m/s} = 150 \text{ mph}$

2.5 (a) Right:  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(b) Wrong: the units do not match (kg/s vs kg/m<sup>3</sup>)

$\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

2.6 (a)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(b)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(c)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(d)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(e)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

(f)  $\frac{100 \text{ g}}{1 \text{ s}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1 \text{ h}}{3600 \text{ s}} = \frac{100}{3600 \times 1000} \text{ kg/s} = (2.78 \times 10^{-5}) \text{ kg/s}$

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**Richard M. Felder, Ronald W. Rousseau**



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