	Activity A:	Get the Gizmo ready: • Use the arrows to create an atom with	two protons.
	Subatomic two neutrons, and two electrons. particles Turn on Show element name.		
stic	on: What are the	properties of protons, neutrons, and electrons	7 Mass number Electical
num	bers surround the	we element symbol and Element notation. Three element symbol: the mass number (A), electrical displayed if the atom is neutral), and the atomic	· '4 LI o
inwe	stigate: Watch how the numbers change as you add or remove particles.		
A.	Which number is	s equal to the number of protons in the atom?	neutrons
В.	How can you calculate the number of neutrons (N) in an atom?		by the number of proton
C.	Which particle (proton, neutron, or electron) has a		
	Positive charge?		electron
	Negative charge	?	proton
	-		-

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