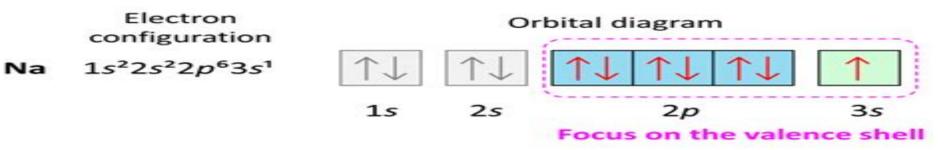
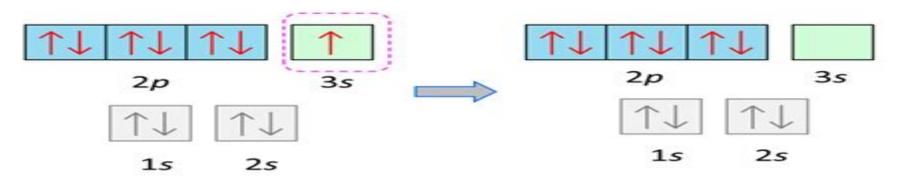
Electron Configuration of Main Group Cations



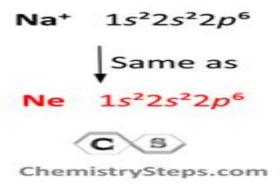
The charge is +1, therefore, one electron is lost:

The electron must be from the outermost valence orbital $-3s^1$



Na $1s^2 2s^2 2p^6 3s^1$

Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are *isoelectronic* with the previous noble gas.



Electron Configurations Of Atoms And Ion Lab

Terry C. Jones

Electron Configurations Of Atoms And Ion Lab:

Electron Configurations Of Atoms And Ion Lab Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Electron Configurations Of Atoms And Ion Lab**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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