

Element/ Ion	Atomic no.	# of $e^-$	Electronic configuration
F	9	9	$1s^2 2s^2 2p^5$
$F^-$	9	10	$1s^2 2s^2 2p^6$
O	8	8	$1s^2 2s^2 2p^4$
$O^{2-}$	8	10	$1s^2 2s^2 2p^6$
Na			
$Na^+$			
Ca			
$Ca^{2+}$			

# Electron Configuration Of Atoms And Ions Lab Answers

**KJ Lindholm-Leary**



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