

Electrochemistry-Ch.9 – PRACTICE TEST/EXAM QUESTIONS

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- ____ 1. In the reaction
$$2HNO_{3(aq)} + Cu_{(s)} + 2H^{+}_{(aq)} \rightarrow 2NO_{2(g)} + Cu^{2+}_{(aq)} + 2H_2O_{(l)}$$
Which of the following statements is correct?
a. HNO_3 is the reducing agent, and Cu is the oxidizing agent
b. HNO_3 is the reducing agent, and H^{+} is the oxidizing agent
c. Cu is the reducing agent, and H^{+} is the oxidizing agent
d. Cu is the reducing agent, and HNO_3 is the oxidizing agent
e. H^{+} is the reducing agent, and Cu is the oxidizing agent
- ____ 2. The oxidation number of chromium (Cr) in the dichromate ion ($Cr_2O_7^{2-}$) is
a. +2
b. +4
c. +6
d. +7
e. -2
- ____ 3. The oxidation number of fluorine (F) in AlF_3 is
a. -3
b. -2
c. -1
d. 0
e. +1
- ____ 4. In $Fe_2(SO_4)_3$ the oxidation numbers of Fe , S , and O respectively are
a. +2, +3, -4
b. +3, +6, -2
c. +2, +4, -8
d. +2, +4, -2
e. +4, +5, -3
- ____ 5. Which substance in the following reaction has undergone oxidation?
$$2MnO_4^{-}(aq) + 5C_2O_4^{2-}(aq) + 16H^{+}(aq) \rightarrow 2Mn^{2+}(aq) + 10CO_{2(g)} + 8H_2O_{(l)}$$

a. MnO_4^{-}
b. $C_2O_4^{2-}$
c. H^{+}
d. Mn^{2+}
e. CO_2
- ____ 6. This question consists of an assertion (statement) in the left-hand column and a reason in the right-hand column.

Assertion

Chlorine will cause iodine to form when chlorine is bubbled through a solution of potassium iodide.

Reason

Chlorine is a stronger oxidizing agent than iodine.

Select one of the following.

- if both assertion and reason are true statements, and the reason is a correct explanation of the assertion
- if both assertion and reason are true statements, and the reason is not a correct explanation of the assertion
- if the assertion is true but the reason is false
- if the assertion is false but the reason is a true statement
- if both assertion and reason are false statements

Electrochemistry 2chapter Test A Answer Key

Neil D. Jespersen, Pamela Kerrigan



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Gupta, 2021-10-17 1 The book is prepared for the problem solving in chemistry 2 It is divided into 5 chapters 3 Each chapter is topically divided into quick theory Immediate Test and Knowledge Confirmation Test 4 At the end of the each chapter cumulative exercises for JEE Main Advanced for practice 5 Acid Test for JEE Mains Advance containing all types of questions asked in JEE A common phrase among JEE Aspirants that chemistry is the most scoring subject but the problems asked in JEE Exams are not directly related but they are based on multiple applications Introducing the all new edition of Problem Physical Chemistry JEE Main Advanced Volume 2 which is designed to develop the use of the concepts of chemistry in solving the diversified problems as asked in JEE The book divides the syllabus into 5 chapters and each chapter has been topically divided in quick theory different types of Solved Examination followed by Immediate Test along with the Topicwise short exercises Knowledge Confirmation Test At the end of each chapter there are separate cumulative exercises for JEE Main Advanced Acid Test for JEE Mains Advance are also provided containing all types of questions asked in JEE Detailed and explanatory solutions provided to all the questions for the better understanding TOC Solid State Solution and Colligative Properties Electrochemistry Chemical Kinetics Surface Chemistry *The Use of Synthetic Environments for Corrosion Testing* P. E. Francis, Thaddeus Shuptrine Lee, 1988 **Biomedical Science Practice** Nessar Ahmed, 2022 The Fundamentals of Biomedical Science series has been written to reflect the challenges of practicing biomedical scientists today It draws together essential basic science with insights into laboratory practice to show how an understanding of the biology of disease is linked to analytical approaches that lead to diagnosis The series reviews the full range of disciplines to which a biomedical scientist may be exposed from microbiology to cytopathology to transfusion science The third edition of Biomedical Science Practice gives a comprehensive overview of key laboratory techniques and professional practical skills with which students will need to be familiar to be successful in a professional biomedical environment The text discusses a broad range of professional skills and concepts such as health and safety considerations personal development and communication and confidentiality The text also explores key experimental and analytical approaches which form the basis of the investigation and diagnosis of clinical conditions Each chapter is supported with engaging clinical case studies written to emphasize the link between theory and practice and a set of end of chapter questions which encourages students to test their knowledge and stretch their understanding The third edition is available for students and institutions to purchase in a variety of formats and is supported by online resources The e book offers a mobile experience and convenient access along with functionality tools navigation features and links that offer extra learning support www.oxfordtextbooks.co.uk/ebooks Online student resources supporting the book include Answers to case study and self check questions Multiple choice questions An interactive Digital Microscope encouraging the exploration of tissue samples Video podcasts including interviews with practicing biomedical scientists and in the lab footage showing biomedical science in practice Online lecturer resources supporting the book include Figures from the book available to download **DAT: Dental Admissions Test: Includes 3**

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The Industrial Chemist and Chemical Manufacturer, 1925

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