Electrochemical Cells and Thermodynamics

17

Shorthand cell designation Temperature (TO) E cell (measured) SG (calculated) K_ (calculated)

1. 9 Pb | CU 20.2 . 492 U -94 KT 14 CL 25

2. 50 11 CU 8000003 . 249 V -48 KT 12,198

3. 50 11 Pb 20.9 . 125 V -16 KFT 5.645

Show calculations for ΔG^* and K_{eq} for an exemplary pair, For Pb | Pb²⁺ | Sn²⁺ | Sn:

Halfycell equation	2+ > -	E half-cell (calculated)		
1. (Pb(s) -) 2.(5n -75n +2e) 3.(5n -75n + 2e)	Ph + 20 /(00	12e -) CU	= 4060	
2 (5n -> 5n + 2e-)	(co", se -)(w)	= -	. 425 V	
3/50 -> 50T + 2e)	(Phoi Loe >Pb)	_ =	.01961	

Effect of Temperature on Cell Potential

Cell designation: E (measured) P511Cu: -08 V	Temperature (°C)	Temperature (K)	AG (calculated)
51 4 Co : .085V	35	- 308 _	-16,405
50 11 Pb0 x60	58	323	-16,217
P51/CU- 1087U -	_60	333	-18,653
50 11 CV: 10814	105	338	-18,097
5 UIL Pb: 1079U	70	_ 371	

 ΔS determined from the slope of a plot of ΔG versus T=3, G=2

ΔH° calculated at 298 K - 7, 652 (show calculations)

Electrochemical Cells And Thermodynamics Lab Answers

Brendan G. Carr

Electrochemical Cells And Thermodynamics Lab Answers:

Reviewing **Electrochemical Cells And Thermodynamics Lab Answers**: Unlocking the Spellbinding Force of Linguistics

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