

8. Given the following data for a fluid, what is its state at 40°C and 3 kPa?

Saturated property table

| $T_s, ^\circ\text{C}$ | $P_s, \text{kPa}$ | $v_{f,s}, \text{m}^3/\text{kg}$ | $v_{g,s}, \text{m}^3/\text{kg}$ | $h_{f,s}, \text{kJ/kg}$ | $h_{g,s}, \text{kJ/kg}$ | $s_{f,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$ | $s_{g,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$ |
|-----------------------|-------------------|---------------------------------|---------------------------------|-------------------------|-------------------------|------------------------------------------------|------------------------------------------------|
| 40                    | 7.38              | .001008                         | 19.52                           | 167.57                  | 2574.3                  | .5725                                          | 8.257                                          |
| 80                    | 47.39             | .001029                         | 3.407                           | 334.9                   | 2643.7                  | 1.1343                                         | 7.5645                                         |
| 120                   | 198.5             | .001060                         | .8919                           | 503.71                  | 2706.3                  | 1.5276                                         | 7.1296                                         |

- (A) Saturated liquid
- (B) Superheated vapor
- (C) Compressed liquid
- (D) Saturated vapor

9. Using the same data of Problem 8, what is the fluid's entropy in kJ/(kgK) at 120°C and 80% quality?

- (A) 1.53
- (B) 6.009
- (C) 7.13
- (D) 28.8

10. A quantity of 55 m<sup>3</sup> of water passes through a heat exchanger and absorbs 2,800,000 kJ. The exit temperature is 95°C. The entrance water temperature in °C is nearest to:

- (A) 49
- (B) 56
- (C) 68
- (D) 83

11. In any non-quasistatic thermodynamic process, the overall entropy of an isolated system will:

- (A) Increase and then decrease
- (B) Decrease and then increase
- (C) Stay the same
- (D) Increase only

12. The temperature-entropy diagram in the figure represents a:

- (A) Rankine cycle with superheated vapor
- (B) Carnot cycle
- (C) Diesel cycle
- (D) Refrigeration cycle



13. Ethane is burned in 150% theoretical air. Assume complete combustion and that the process takes place in atmospheric air. The air/fuel ratio in 150% theoretical air is most nearly:

- (A) 32 kg air/kg fuel
- (B) 28 kg air/kg fuel
- (C) 24 kg air/kg fuel
- (D) 16 kg air/kg fuel

14. There are 3 kg of air in a rigid container at 250 kPa and 50°C. The gas constant for air is 0.287 kJ/(kgK). The volume of the container, in m<sup>3</sup>, is nearest to:

- (A) 2.2
- (B) 1.1
- (C) 2.8
- (D) 3.1

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**Padhraic Smyth**



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