

| Page 3 | Mark Scheme Chemistry - June 2004 | Syllabus 0620 | Paper 02 |
|--------|--------------------------------------|------------------|-------------|
|--------|--------------------------------------|------------------|-------------|

- 4** (a) ethanol - solvent
ethene - polymer
bitumen - roads [3]
- (b) ethanol [1]
- (c) (i) C [1]
(ii) A [1]
(iii) B [1]
(iv) D [1]
- (d) (i) (compound) containing only carbon and hydrogen
NOT: it contains carbon and hydrogen [1]
(ii) has only single bonds/ has general formula C_nH_{2n+2}
NOT: it is saturated [1]

Total = 10

- 5** (a) chlorine, argon, potassium, bromine, iodine [1]
ALLOW: symbols
- (b) chlorine, potassium, argon, bromine, iodine [1]
ALLOW: symbols
- (c) 2nd box down ticked [1]
- (d) chlorine, bromine, iodine (all 3 needed) [1]
ALLOW: symbols
- (e) (i) potassium/K [1]
(ii) argon/Ar [1]
- (f) 1st and 4th boxes ticked (1 mark each) [2]
- (g) (i) high (boiling point) [1]
(ii) conducts/is high [1]
- (h) potassium loses an/one electron/loses outer shell [1]
chlorine gains an/one electron/outer shell becomes complete [1]
ALLOW: (for 1 mark) potassium loses two electrons + chlorine gains two electrons
ALLOW: e.g. 2.8.8.1 → 2.8.8 for first mark
Any indication of sharing electrons = 0

Total = 12

Ms 0620 6m J 10

Julia Schneider



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1. Understanding the eBook Ms 0620 6m J 10
 - The Rise of Digital Reading Ms 0620 6m J 10
 - Advantages of eBooks Over Traditional Books
2. Identifying Ms 0620 6m J 10
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ms 0620 6m J 10
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ms 0620 6m J 10
 - Personalized Recommendations
 - Ms 0620 6m J 10 User Reviews and Ratings

- Ms 0620 6m J 10 and Bestseller Lists
- 5. Accessing Ms 0620 6m J 10 Free and Paid eBooks
 - Ms 0620 6m J 10 Public Domain eBooks
 - Ms 0620 6m J 10 eBook Subscription Services
 - Ms 0620 6m J 10 Budget-Friendly Options
- 6. Navigating Ms 0620 6m J 10 eBook Formats
 - ePub, PDF, MOBI, and More
 - Ms 0620 6m J 10 Compatibility with Devices
 - Ms 0620 6m J 10 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ms 0620 6m J 10
 - Highlighting and Note-Taking Ms 0620 6m J 10
 - Interactive Elements Ms 0620 6m J 10
- 8. Staying Engaged with Ms 0620 6m J 10
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ms 0620 6m J 10
- 9. Balancing eBooks and Physical Books Ms 0620 6m J 10
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ms 0620 6m J 10
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ms 0620 6m J 10
 - Setting Reading Goals Ms 0620 6m J 10
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ms 0620 6m J 10
 - Fact-Checking eBook Content of Ms 0620 6m J 10
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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