



ACIDS & BASES

Acids

Acids are sour, conduct electricity, and react with metals to produce hydrogen gas. Lemon contains acid.

Bases

Bases are bitter, feel slippery, and react with acids to produce salts. Baking soda is a base.

Neutral Substances

A neutral substance is neither an acid or base. Pure water, for example, is a neutral substance.

Indicators

It changes color to indicate whether a solution is an acid, a base, or neutral. It does not tell how strong the acid or base is.

Acidic Region	Neutral	Basic Region
1 (Strong Acid)	7 (Neutral)	14 (Strong Base)
2 (Weak Acid)		
3 (Weak Acid)		
4 (Weak Acid)		
5 (Weak Acid)		
6 (Weak Acid)		
7 (Neutral)		
8 (Weak Base)		
9 (Weak Base)		
10 (Weak Base)		
11 (Weak Base)		
12 (Weak Base)		
13 (Weak Base)		
14 (Strong Base)		

Acid-Base Indicators

Indicator	Color in Acid	Color in Base
Phenolphthalein	Colorless	Pink
Universal Indicator	Red	Blue
Methyl Orange	Red	Yellow
Blue Litmus	Red	Blue
Red Litmus	Red	Blue

STATES OF MATTER

All matter is made of particles. The state of matter is determined by the speed of the particles and the strength of the attractive forces between them. Energy must be supplied or removed in order for substances to change from one state to another.

SOLID

Particles have definite shape and volume. The particles in a solid are packed closely together because they don't move fast enough to overcome the attractive forces between them.

LIQUID

Liquids have definite volume but take the shape of the container holding them. The particles in a liquid move just fast enough to overcome some of the attractive forces between them and slide past one another.

GAS

Gases have no definite shape or volume. The particles in a gas move fast enough to overcome almost all attractive forces between them and move independently of one another. Gases can expand to fill the container they are in. Gases can also be compressed.

PLASMA

Plasma forms when the temperature of a gas is extremely high and the particles are ionized and move independently of one another. Plasma can conduct electricity, emit light, and is affected by magnetic fields. Plasma occurs naturally in lightning.

PHYSICAL & CHEMICAL CHANGES

PHYSICAL CHANGES

During a physical change, the identity of the substance does not change. Only the form or appearance of the substance changes. No new substance is formed.

Examples of Physical Changes:

- Melting:** Change from solid to liquid.
- Freezing:** Change from liquid to solid.
- Boiling:** Change from liquid to gas.
- Condensation:** Change from gas to liquid.
- Sublimation:** Change from solid to gas.
- Deposition:** Change from gas to solid.

Most physical changes can be reversed or undone.

CHEMICAL CHANGES

During a chemical change, a substance's molecules break apart and are rearranged into new substances. The identity of the substance changes. A new substance is formed.

Signs that a Chemical Change Has Occurred:

- Color Change
- Formation of a Gas
- Formation of a Precipitate
- Change in Temperature

Chemical changes are irreversible or difficult to reverse.

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