

### HYDROGEN SULFIDE (H₂S) AWARENESS PROGRAM

#### PURPOSE

This safety awareness program is intended to provide information to Winger Companies, herein referred to as Winger, employees regarding the potential health effects of Hydrogen Sulfide (H<sub>2</sub>S). Our goal is to ensure our employees have the knowledge they need to work safely and adequate measures can be taken to limit exposures through safe work procedures in the workplace.

#### WHAT IS HYDROGEN SULFIDE (H2S)?

Hydrogen sulfide is the chemical compound with the formula H<sub>2</sub>S. It is slightly (20%) heavier than air, has a tendency to settle in low-laying areas, and is readily dispersed by wind movements or currents. H<sub>2</sub>S dissolves in water forming a weak acid (hydro sulfurous acid). H<sub>2</sub>S will be released when in water when agitated making it a dangerous hidden hazard. A mixture of H<sub>2</sub>S and air is explosive. When ignition occurs, the combustion produces irritants and toxic gases, including sulfur dioxide (SO2) which can have an irritating effect on the eyes and lungs and can be fatal. H<sub>2</sub>S attacks most metals, especially in the presence of water, forming sulfides that are usually insoluble precipitates. It is also very corrosive to plastics and tissue.

Hydrogen sulfide is a colorless, very poisonous, flammable, extremely hazardous gas with the characteristic odor of "rotten egg". It occurs naturally in crude petroleum, natural gas, blogas, LPG, volcanic gases, some well waters, and hot springs. Natural gas can contain up to 90%. Hydrogen sulfide often results from the bacterial breakdown of organic matter in the absence of oxygen, such as in swamps and human and animal sewers (e.g., sewage); this process is commonly known as anaerobic digestion. Bacteria found in your mouth and gastrointestinal tract produce hydrogen sulfide from bacteria decomposing materials that contain vegetable or animal proteins. The human body produces small amounts of H<sub>2</sub>S and uses it as a signaling molecule.

Industrial activities that can produce the gas include petroleum/natural gas drilling and refining, wastewater treatment, coke ovens, food processing, tanneries, and kraft paper mills. Hydrogen sulfide can also exist as a liquid compressed gas.

Other names for hydrogen sulfide:

- Dihydrogen monosulfide
- Dihydrogen sulfide
- Hydrosulfuric acid
- Manure gas
- Rotten egg gas
- Sewer gas
- Sour gas
- Stink damp
- Sulfane
- Sulfur Hydride
- Sulfurated hydrogen
- Sulfur hydride
- Swamp das







# **Hydrogen Sulfide Rescue Manual**

Jessica J Manson

**Hydrogen Sulfide Rescue Manual:** 

This book delves into Hydrogen Sulfide Rescue Manual. Hydrogen Sulfide Rescue Manual is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and indepth insights into Hydrogen Sulfide Rescue Manual, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Hydrogen Sulfide Rescue Manual
  - Chapter 2: Essential Elements of Hydrogen Sulfide Rescue Manual
  - o Chapter 3: Hydrogen Sulfide Rescue Manual in Everyday Life
  - Chapter 4: Hydrogen Sulfide Rescue Manual in Specific Contexts
  - ∘ Chapter 5: Conclusion
- 2. In chapter 1, the author will provide an overview of Hydrogen Sulfide Rescue Manual. This chapter will explore what Hydrogen Sulfide Rescue Manual is, why Hydrogen Sulfide Rescue Manual is vital, and how to effectively learn about Hydrogen Sulfide Rescue Manual.
- 3. In chapter 2, the author will delve into the foundational concepts of Hydrogen Sulfide Rescue Manual. This chapter will elucidate the essential principles that must be understood to grasp Hydrogen Sulfide Rescue Manual in its entirety.
- 4. In chapter 3, this book will examine the practical applications of Hydrogen Sulfide Rescue Manual in daily life. The third chapter will showcase real-world examples of how Hydrogen Sulfide Rescue Manual can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of Hydrogen Sulfide Rescue Manual in specific contexts. This chapter will explore how Hydrogen Sulfide Rescue Manual is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Hydrogen Sulfide Rescue Manual. The final chapter will summarize the key points that have been discussed throughout the book.
  - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Hydrogen Sulfide Rescue Manual.

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