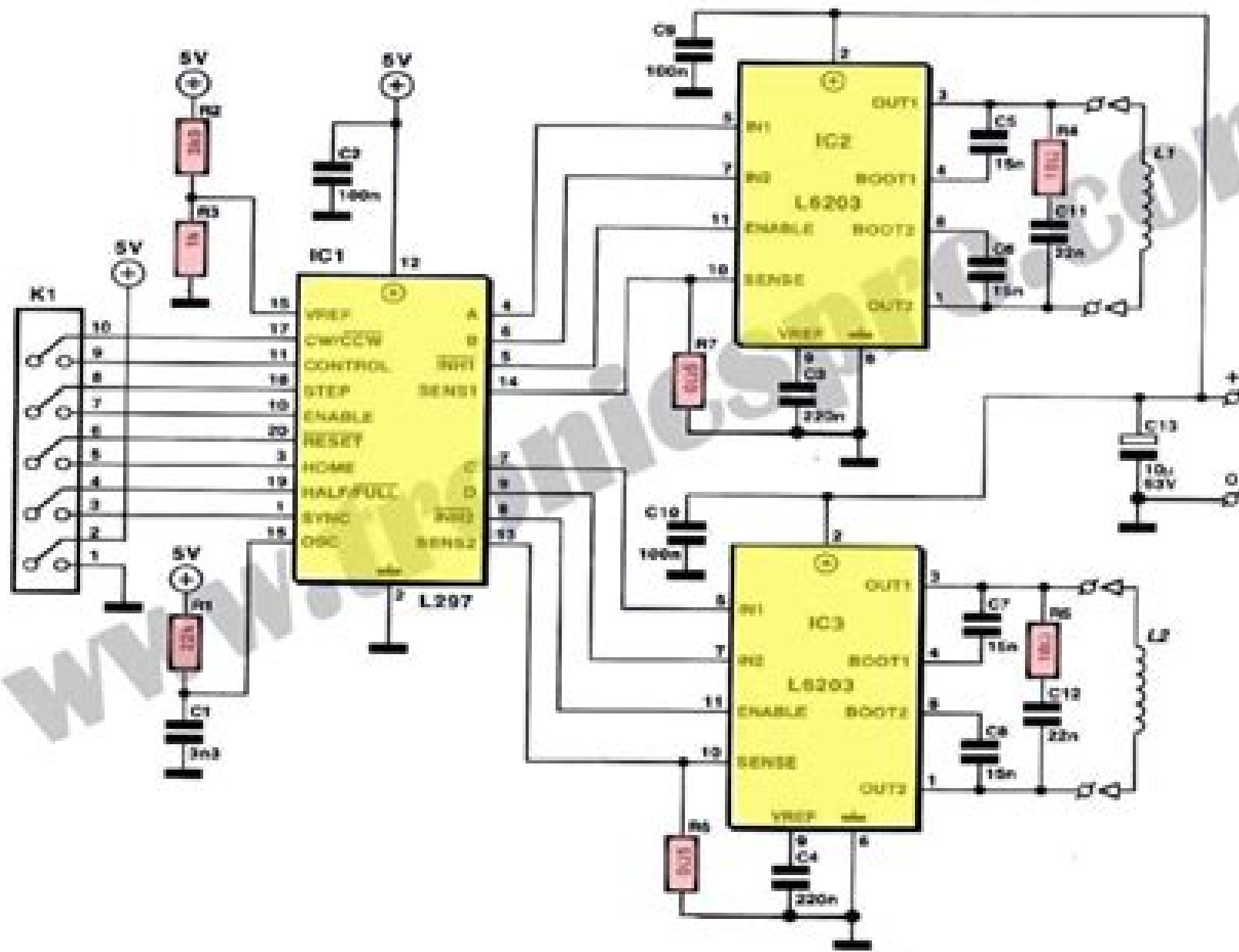


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web in the well known formol titration method of sørensen 22 neutralised formaldehyde is added to the solution of the amino acid or other compound containing NH_2 and standard alkali run in until the production of a red colour with phenolphthalein

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web the formol titration as described by sørensen 1 has been most useful in the determination of amino acids and especially in following the course of hydrolysis of proteins

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web experiment 1 estimation of glycine by sorenson s formol titration aim to estimate the amount of amino acid present in the given solution principle the acid group present in the glycine can be titrated with NaOH