

Half Modern Chemistry Review

CHAPTER 11: CHEMICAL BONDING

This half-modern program concludes that level (that vast 50%) of that information for this chapter is dead.

Focus on this content, but make sure to review class notes, activities, materials, experiments, etc.

If you study this document and DISCOVERED what, you should at least be able to PLACE that item.

Now that items will be recall, examples, and/or application of this content. Now

OUTCOMES

- Communicate with precision the understanding/properties learned (p2 c2)
- Communicate understanding learned to teachers and peers (p2 c2)
- EL-2 Give the Lewis dot notation, illustrate the placement bond on the drawing of molecules (p2 c2)
- EL-2 List important physical properties of some compounds; name molecules/compounds in some (p2 c2, p2 c2)

EL-1: INTRODUCTION TO CHEMICAL BONDING

Introduction

- chemical bond → the attractive force that binds atoms or ions together
- ionic bonding → a force that attracts atoms with some ions has developed, which demonstrates a mutual attraction between them
- covalent bonding → a bond formed when atoms share some or several pairs of valence electrons
- molecular compound bond → a compound bond is formed by sharing electrons and usually represents the basic chemical atoms
- gas → substance, a molecule or atoms that particles are separated by large and irregular spaces
- gas compound bond → a chemical bond in which a gas and molecules represent by that atoms in bond space occupy the space

Chapter Highlights

- Matter atoms are chemically bonded by ionic atoms. The most major types of chemical bonding are ionic, covalent, and metallic
- In general, atoms of metals, bond covalently with atoms of nonmetals, atoms of metals bond covalently with each other, and atoms of nonmetals bond covalently with each other

EL-2: COVALENT BONDING AND MOLECULAR COMPOUNDS

Introduction

- molecule → a group of atoms that are held together by chemical bonds; a molecule is the smallest unit of matter that can exist by itself and retains all of its characteristics
- molecular compound → a chemical compound whose simplest unit, and represents chemical formula
- chemical formula → a combination of chemical symbols and numbers to represent a substance
- molecular formula → a chemical formula that shows the number and kinds of atoms in a molecule, but not the arrangement of the atoms

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