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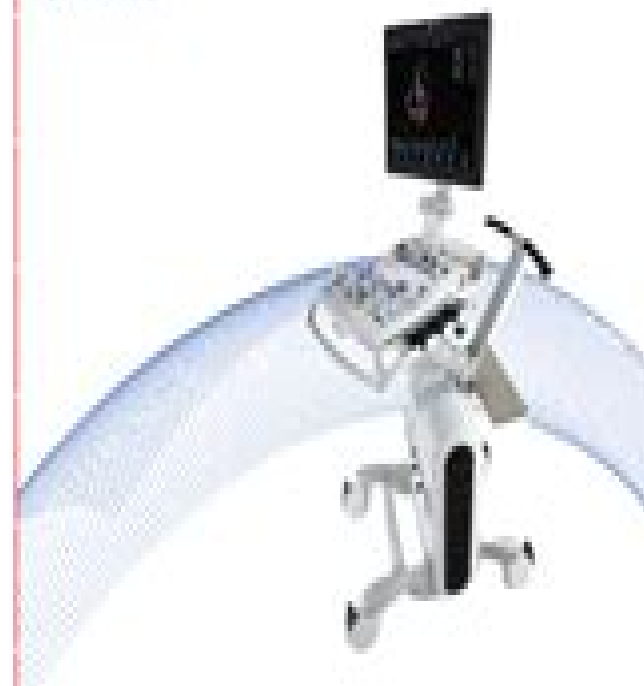
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web jan 10 2023 introduction in this lab we will apply the concepts of reaction rate order of reaction and catalyst our example reaction is the iodination of acetone catalyzed by hydrochloric acid the reaction is zero order in iodine you will figure out the order in acetone and the order in hcl and the rate constant

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web the iodination of acetone is convenient because the color of the iodine reactant is readily observable and the change in concentration of 4 fthe iodine can be easily monitored by our spectrophotometer and the reaction proceeds quickly enough for multiple trials but slowly enough to obtain measurements

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