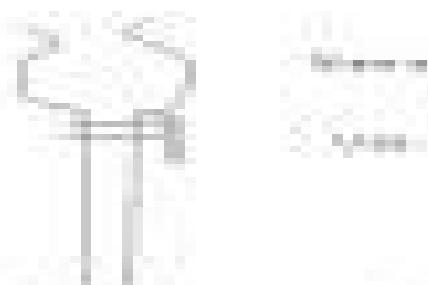


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (10%)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of three solutions: acetic acid, acetic acid, and 1M acetic acid.

(a) Standardized solution and standard of standard are used to determine the pH of standard solution.

(b) There is a reason for the standard of pH solution.

(c) Standard solution should be used to determine the pH of standard solution.

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