

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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ML Yell



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Table of Contents Energy Transfer Concept Review Holt Science Spectrum

1. Understanding the eBook Energy Transfer Concept Review Holt Science Spectrum
 - The Rise of Digital Reading Energy Transfer Concept Review Holt Science Spectrum
 - Advantages of eBooks Over Traditional Books
2. Identifying Energy Transfer Concept Review Holt Science Spectrum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Energy Transfer Concept Review Holt Science Spectrum
 - User-Friendly Interface
4. Exploring eBook Recommendations from Energy Transfer Concept Review Holt Science Spectrum
 - Personalized Recommendations
 - Energy Transfer Concept Review Holt Science Spectrum User Reviews and Ratings
 - Energy Transfer Concept Review Holt Science Spectrum and Bestseller Lists
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 - ePub, PDF, MOBI, and More
 - Energy Transfer Concept Review Holt Science Spectrum Compatibility with Devices
 - Energy Transfer Concept Review Holt Science Spectrum Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Energy Transfer Concept Review Holt Science Spectrum
 - Highlighting and Note-Taking Energy Transfer Concept Review Holt Science Spectrum
 - Interactive Elements Energy Transfer Concept Review Holt Science Spectrum
- 8. Staying Engaged with Energy Transfer Concept Review Holt Science Spectrum
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Energy Transfer Concept Review Holt Science Spectrum
- 9. Balancing eBooks and Physical Books Energy Transfer Concept Review Holt Science Spectrum
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Energy Transfer Concept Review Holt Science Spectrum
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Energy Transfer Concept Review Holt Science Spectrum
 - Setting Reading Goals Energy Transfer Concept Review Holt Science Spectrum
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Energy Transfer Concept Review Holt Science Spectrum
 - Fact-Checking eBook Content of Energy Transfer Concept Review Holt Science Spectrum
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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