

CHAPTER 15 STUDY GUIDE

Energy and Chemical Change

Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- _____ 1. *Energy* is the ability to do work or produce heat.
- _____ 2. The law of conservation of energy states that energy *can be* created and destroyed.
- _____ 3. Chemical potential energy is energy stored in a substance because of its *composition*.
- _____ 4. *Heat* is a form of energy that flows from a warmer object to a cooler object.
- _____ 5. A calorie is the amount of energy required to raise the temperature of *one gram* of pure water by one degree Celsius.
- _____ 6. A *calorie* is the SI unit of heat and energy.
- _____ 7. The *specific* heat of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
- _____ 8. *Kinetic energy* is energy of motion.
- _____ 9. Chemicals participating in a chemical reaction contain only *potential energy*.
- _____ 10. One nutritional Calorie is equal to *100 calories*.
- _____ 11. One calorie equals *4.184 joules*.
- _____ 12. When a fuel is burned, some of its *chemical potential energy* is lost as heat.
- _____ 13. To convert kilojoules to joules, *divide* the number of kilojoules by 1000 joules/1 kilojoule.

Answer the following question. Show all your work.

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

Energy And Chemical Change Study Guide Answer

Thomas Griffiths



Energy And Chemical Change Study Guide Answer:

Whispering the Strategies of Language: An Mental Journey through **Energy And Chemical Change Study Guide Answer**

In a digitally-driven earth wherever monitors reign supreme and immediate interaction drowns out the subtleties of language, the profound techniques and emotional subtleties hidden within phrases usually get unheard. However, situated within the pages of **Energy And Chemical Change Study Guide Answer** a fascinating literary prize blinking with fresh emotions, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, this enchanting opus encourages viewers on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the material of each word. Within the mental depths of the moving review, we will embark upon a heartfelt exploration of the book is core styles, dissect their fascinating publishing design, and succumb to the effective resonance it evokes strong within the recesses of readers hearts.

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