

## Energy Changes in Chemical reactions for KS3 Science - Worksheet (Answer)

- In an exothermic reaction does the temperature go up or down?  
The temperature will increase
  - In an endothermic reaction does the temperature go up or down?  
The temperature will decrease
  - Name two examples of exothermic reactions.  
Respiration & combustion
  - Name two examples of endothermic reactions.  
Photosynthesis & melting / decomposition
  - Complete the correct sentence.  
The words between the arrows of the reaction equation below mean that this is an exothermic reaction. Therefore, the words are likely to mean the arrow of the equation points to the right, this is an exothermic reaction, correct?
  - Use the table to answer the question.
- | Reaction | Starting temperature / °C | Final temperature / °C |
|----------|---------------------------|------------------------|
| A        | 20                        | 30                     |
| B        | 20                        | 10                     |
| C        | 20                        | 20                     |
- Which reaction was exothermic? exothermic is exothermic, correct? yes or no?  
a. Exothermic temperature increases  
b. Exothermic temperature decreases  
c. Exothermic temperature increases
  - Which reaction has the biggest energy change?  
a. biggest temperature change
  - In an exothermic reaction, is enthalpy change positive or negative?  
negative
  - In an endothermic reaction, is enthalpy change positive or negative?  
positive
  - When hydrochloric acid reacts with ammonium hydroxide to create the temperature goes up  
acid + alkali → alkali + salt  
acid + alkali  
Complete the energy profile diagram and state whether the reaction is exothermic or endothermic, explain your answer.

# Energy And Chemical Change Answers

**Xiaolong Qi**



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