

Name	Class	Date given
		Date due in

Chemical Reactions

1. Tick the correct statement below. [1]

- In a chemical reaction...
- ☐ atoms and mass are lost.
- ☐ atoms and mass are gained.
- ☐ atoms are rearranged and mass is conserved.

Complete the equations below. [4]

2a. iron + _____ → iron oxide

Fe + O₂ → _____

2b. copper carbonate → copper oxide + _____

_____ → CuO + CO₂

3a. Circle the type of reaction that is shown in question 2a. [1]

oxidation combustion thermal decomposition

3b. Circle the type of reaction that is shown in question 2b. [1]

oxidation combustion thermal decomposition

4. Match the sentences by writing the correct letter for the end part of each sentence and in the corresponding beginning part. [2]

- a. An exothermic reaction happens when... _____
- b. An endothermic reaction happens when... _____
- c. A catalyst is a substance... _____

- i. ...speeds up a reaction
- ii. ...is unaffected by the surroundings
- iii. ...is unaffected by the surroundings

Learning objectives met if achieved

2a	iron reacts in a chemical reaction to	
2b	iron oxide, copper carbonate, carbon dioxide and copper sulphate	
3a	an exothermic reaction type of chemical reaction	
3b	an endothermic reaction type of chemical reaction / a catalyst	



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Version 1.0



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oxidation combustion thermal decomposition

3b. Circle the type of reaction that is shown in question 2b. [1]

oxidation combustion thermal decomposition

4. Match the sentences by writing the correct letter for the end corresponding beginning part. [2]

- a. An exothermic reaction happens when... _____

ink saving Eco

End Of Unit Test Chemical Reactions Ks3

KJ Lindholm-Leary



End Of Unit Test Chemical Reactions Ks3:

Spotlight Science Keith Johnson, Sue Adamson, Gareth Williams, 2000 Topic Outlines show parts of the PoS to be covered the relationship of the topic to aspects of KS2 and KS4 and warn of equipment that may need special preparation time in advance Topic Maps are provided for students Lesson Notes relating to each double page spread in the students book offer objectives ideas for each lesson detailed references to the PoS level descriptions safety points with references to CLEAPPS HAZCARDS ICT support cross curricular links and equipment lists Answers to all questions in the students book are also provided Additional support material provide Homework Sheets Help and Extension Sheets to optimise differentiation Sc1 Sc1 Skill Sheets Thinking about activities to improve integration of CASE activities with Spotlight Science Revision Quizzes and Checklists etc Extra Help Sheets for each topic extend the range of support for Sc1 and Sc2 4 Challenge Sheets for each topic provide a variety of enrichment activities for more able students They consist of a variety of challenging activities which will present students with opportunities to develop problem solving thinking presentational and interpersonal skills Technician s Cards include help to prepare lessons equipment requirements and CLEAPPS HAZCARD references For more information visit the website at www.spotlightscience.co.uk **The Times Index** ,2008 Indexes the Times and its supplements **Passing the North Carolina End of Course Test for Chemistry** Liz Thompson, 2007 **Chemistry** Gavin Whittaker, 2001

End Of Unit Test Chemical Reactions Ks3 Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**End Of Unit Test Chemical Reactions Ks3**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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