

Empirical/Molecular Formula

Name: _____

Date: _____

1. What is an empirical formula? How is it different from a molecular formula?

The empirical formula is the lowest whole number ratio of atoms in a compound. The molecular formula shows the true number of atoms in the molecule and is a whole number multiple of the empirical formula.

2. State the empirical formula for each of the following.



3. A sample of a compound is 50.0% sulfur and 50.0% oxygen by mass. What is the empirical formula?

$$S: \frac{50.0 \text{ g S}}{32.07 \text{ g S}} \times \frac{1 \text{ mol S}}{1} = \frac{1.559 \text{ mol S}}{1.559 \text{ mol}} = 1 \text{ S}$$

EF: SO_2

$$O: \frac{50.0 \text{ g O}}{16.00 \text{ g O}} \times \frac{1 \text{ mol O}}{1} = \frac{3.125 \text{ mol O}}{1.559 \text{ mol}} = 2 \text{ O}$$

4. A sample of a compound is 52.9% aluminum. The rest is oxygen. What is the empirical formula?

$$Al: \frac{52.9 \text{ g Al}}{26.98 \text{ g Al}} \times \frac{1 \text{ mol Al}}{1} = \frac{1.961 \text{ mol Al}}{1.961 \text{ mol Al}} = 1 \text{ Al} \times 2 = 2$$

EF: Al_2O_3

$$O: \frac{47.1 \text{ g O}}{16.00 \text{ g O}} \times \frac{1 \text{ mol O}}{1} = \frac{2.944 \text{ mol O}}{1.961 \text{ mol}} = 1.5 \times 2 = 3 \text{ O}$$

Ratio must be a whole

5. A compound is 48.65% carbon, 8.11 % hydrogen and 43.24 % oxygen. Determine the empirical formula.

4.05 mol carbon, 8.11 mol H, 2.70 mol O



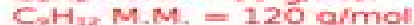
1. Divide by the smallest # to get = $C_{1.5}H_3O_1$

2. Multiply to get a whole number ratio (x2) = $C_3H_6O_2$

6. Vinegar is a dilute solution of acetic acid. The molar mass of acetic acid is 60.0 g/mol and it has an empirical formula of CH_2O . What is the molecular formula of acetic acid?



7. A compound has an empirical formula of C_3H_4 . Which of the following are possible molar masses of the compound? 20 g/mol, 55 g/mol, 80 g/mol, 120 g/mol.



Empirical And Molecular Formula Show Work

**Barron's Educational Series, Gabrielle
I. Edwards, Cynthia Pfirrmann**

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substances with desired properties This knowledge is instrumental in various careers including those of doctors engineers pharmacists veterinarians dentists pharmacologists and chemists Thus a solid grasp of organic chemistry is essential for success in these fields Despite its importance organic chemistry is often perceived as challenging This perception raises questions such as How should one start learning organic chemistry What should be studied and How can one effectively remember chemical reactions This book aims to address these concerns by offering a comprehensive guide that simplifies the study of Pharmaceutical Organic Chemistry Instead of rote memorization this book encourages understanding the subject conceptually It is designed to make learning organic chemistry engaging and enjoyable *e-O-Level Chemistry Learning Through Diagrams* Dr. Norbani Abdullah,2012-03-14 You will find this book interesting Chemistry concepts presented in a diagrammatic form Specially written to ease learning and to stimulate interest in Chemistry this book will help students in acquiring and reinforcing Chemistry concepts and especially the difficult ones more easily and effectively This book makes learning easier through the following features Learning Outcomes Learning outcomes on the header point out the concepts that you should focus on in the process of learning Important Concepts and Key Terms The important concepts and key terms are presented clearly in simple language Further explanations linked to the diagrams help you better understand the concepts Interesting Visuals Visual aids such as concept maps flow charts and annotated diagrams are integrated to make the concepts easier to understand and remember Real life Examples These examples show real life application of concepts and explain the inquiries on the phenomena that happen in our everyday lives Worked Examples Step by step worked examples help to reinforce your skills in solving problems Instant Facts These are extra information that can help you acquire a more in depth understanding of the topic under discussion This book complements the school curriculum and will certainly help in your preparation for the examinations **Self Study Guide for PVT 2022** Arihant Experts,2021-09-02 1 All India Pre Veterinary Test Entrance Examination is prepared for the entrance of the VET 2 The Guide is divided into 4 main sections 3 Complete Study Material as per prescribed syllabus Pattern by AIPVT 4 Previous Years Solved Papers for practice 5 Division of chapters strictly based on the latest syllabus 6 Step by step guidance is provided for better understanding of the concepts To succeed in the AIPVT Examination grab your copies of Self Study Guide PVT All India Pre Veterinary Test a revised edition that has been prepared exactly on the lines of pattern Level and syllabi of the exam Its approach has been kept simple and lucid presented in a Step by Step manner for complete grasp of the content This guide divides the whole syllabus into 4 major categories and every chapter is provided with ample exercises for practice Lastly Previous Years Papers are incorporated to make students familiar with exact examination pattern and trends Enough practice done through this book students will score high with good ranking TOC AIPVT Solved Paper 2021 2018 Physics Chemistry Botany Appendix **Chemistry for the IB Diploma Coursebook with Free Online Material** Steve Owen,Peter Hoeben,Mark Headlee,2014-03-13 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for

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