

Activity A:

Subatomic particles

Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).



2. **Investigate:** Watch how the numbers change as you add or remove particles.

- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons (N) in an atom?
- C. Which particle (proton, neutron, or electron) has a
Positive charge?
Negative charge?

neutrons
by the number of protons

electron
proton

3. **Observe:** Turn on **Element notation** and **Show element symbol**. Turn on **Show element name**. Turn on **Show mass number**, **Show atomic number**, and **Show electrical charge**. Turn on **Show element symbol**. Turn on **Show element name**. Turn on **Show mass number**, **Show atomic number**, and **Show electrical charge**. Turn on **Show element symbol**. Turn on **Show element name**.

- A. What is the mass number of helium?
B. What is the atomic number of helium?
C. What is the electrical charge of helium?

4. **Observe:** Turn on **Show element name** and **Show element symbol**. Turn on **Show mass number**, **Show atomic number**, and **Show electrical charge**. Turn on **Show element symbol**. Turn on **Show element name**. Turn on **Show mass number**, **Show atomic number**, and **Show electrical charge**. Turn on **Show element symbol**. Turn on **Show element name**.

Element Builder Gizmo Answers Key

A Gutmann



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Table of Contents Element Builder Gizmo Answers Key

1. Understanding the eBook Element Builder Gizmo Answers Key
 - The Rise of Digital Reading Element Builder Gizmo Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Element Builder Gizmo Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Element Builder Gizmo Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Element Builder Gizmo Answers Key
 - Personalized Recommendations
 - Element Builder Gizmo Answers Key User Reviews and Ratings
 - Element Builder Gizmo Answers Key and Bestseller Lists

5. Accessing Element Builder Gizmo Answers Key Free and Paid eBooks
 - Element Builder Gizmo Answers Key Public Domain eBooks
 - Element Builder Gizmo Answers Key eBook Subscription Services
 - Element Builder Gizmo Answers Key Budget-Friendly Options
6. Navigating Element Builder Gizmo Answers Key eBook Formats
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 - Element Builder Gizmo Answers Key Compatibility with Devices
 - Element Builder Gizmo Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Element Builder Gizmo Answers Key
 - Highlighting and Note-Taking Element Builder Gizmo Answers Key
 - Interactive Elements Element Builder Gizmo Answers Key
8. Staying Engaged with Element Builder Gizmo Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Element Builder Gizmo Answers Key
9. Balancing eBooks and Physical Books Element Builder Gizmo Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Element Builder Gizmo Answers Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Element Builder Gizmo Answers Key
 - Setting Reading Goals Element Builder Gizmo Answers Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Element Builder Gizmo Answers Key
 - Fact-Checking eBook Content of Element Builder Gizmo Answers Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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