CHAPTER 4 REVIEW

Arrangement of Electrons in Atoms

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

- In what way does the photoelectric effect support the particle theory of light?
 In order for an electron to be ejected from a metal surface, the electron must be struck by a single photon with at least the minimum energy needed to knock the electron loose.
- 2. What is the difference between the ground state and the excited state of an atom?
 The ground state is the lowest energy state of the atom. When the atom absorbs energy, it can move to a higher energy state, or excited state.
- 3. Under what circumstances can an atom emit a photon?

 A photon is emitted when an atom moves from an excited

A photon is emitted when an atom moves from an excited state to its ground state or to a lower-energy excited state.

4. How can the energy levels of the atom be determined by measuring the light emitted from an atom?

When an atom loses energy, it falls from a higher energy state to a lower energy state. The frequency of the emitted light, observed in an element's line-emission spectrum, may be measured. The energy of each transition is calculated using the equation E = hv, where v is the frequency of each of the lines in the element's line-emission spectrum. From the analysis of these results, the energy levels of an atom atom of each element may be determined.

5. Why does electromagnetic radiation in the ultraviolet region represent a larger energy transition than does radiation in the infrared region?

Energy is proportional to frequency, and ultraviolet radiation has a higher frequency than infrared radiation. To produce ultraviolet radiation, electrons must drop to lower energy levels than they do to produce infrared radiation.

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Electrons In Atoms Chapter 5 Worksheet Answers

G Thomas

Electrons In Atoms Chapter 5 Worksheet Answers:

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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Strengthening Physical Science Skills for Middle & Upper Grades, Grades 6 - 12 Shireman, 2009-02-16 Develop interest and confidence in advanced science by building science vocabulary and math skills while exploring physical science concepts In Strengthening Physical Science Skills topics include matter gravity density motion simple machines electricity light and more It also includes a CD ROM with interactive exercises that are automatically scored and printed plus printable worksheets and reading activities It also supports NSE standards Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources **Prentice Hall Physical Science Concepts in Action Program Planner National** Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Learning Elementary** Chemistry for Class 7 (A.Y. 2023-24)Onward Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a

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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 ,

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