

At anode: At cathode: $2 \text{ H}_2\text{O (I)} \longrightarrow \text{O}_2(g) + 4 \text{ H}^{\oplus} + 4 \text{ e}^{\ominus} \qquad 2 \text{ H}^{\oplus}(\text{aq.}) + 2 \text{ e}^{\ominus} \longrightarrow \text{H}_2(g)$

Electrolysis Chapter For Class 10

M Walker

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