

Recitation Worksheet Six

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1. If 0.50 moles of nitrogen gas reacts with 0.50 moles of oxygen gas, how many molecules of nitrogen dioxide are produced?

3.0E23

molecules



$$0.50 \text{ mol } N_2 \times \frac{2 \text{ mol } NO_2}{1 \text{ mol } N_2} \times \frac{6.022 \times 10^{23} \text{ molecules}}{1 \text{ mol } NO_2} = 6.022 \times 10^{23} \text{ molecules } NO_2$$

limiting reagent

$$0.50 \text{ mol } O_2 \times \frac{2 \text{ mol } NO_2}{2 \text{ mol } O_2} \times \frac{6.022 \times 10^{23} \text{ molecules}}{1 \text{ mol}} = 3.011 \times 10^{23} \text{ molecules } NO_2$$

Limiting Reagent Section Review Key

David Baud



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