Limiting Reagent & Percent Yield Practice Worksheet

- When copper (II) chloride reacts with sodium nitrate, copper (II) nitrate and sodium chloride are formed.
 - a. Write the balanced equation for the reaction given above:

 $CuCl_2 + NaNO_3 \rightarrow Cu(NO_3)_2 + NaCl$ $CuCl_2 + 2 NaNO_3 \rightarrow Cu(NO_3)_2 + 2 NaCl$

b. If 15 grams of copper (II) chloride react with 20 grams of sodium nitrate, how much sodium chloride can be formed?

15 g CuCl₂ x 1 mol CuCl₂ x 2 mol NaCl x 58.443 g NaCl = 13 g NaCl 134.362g CuCl₂ 1 mol CuCl₂ 1 mol NaCl

20 g NaNO₃ x <u>1 mol NaNO₃</u> x <u>2 mol NaCl</u> x <u>58.443 g NaCl</u> = 13.8 g NaCl 84.997g NaNO₃ 2 mol NaNO₃ 1 mol NaCl

Only 13 g can be formed

- c. What is the limiting reagent for the reaction in #2? copper (II) chloride
- d. How many grams of copper (II) nitrate is formed?

e. How much of the excess reagent is left over in this reaction?

15g CuCl₂ x 1 mol CuCl₂ x 2 mol NaNO₃ x 84,997g NaNO₃ = 19 g NaNO₃ 134,452g CuCl₂ 1 mol CuCl₂ 1 mol NaNO₃

 $20 \text{ g NaNO}_3 - 19 \text{ g NaNO}_3 = 1 \text{ g NaNO}_3$

f. If 11.3 grams of sodium chloride are formed in the reaction described in problem #2, what is the percent yield of this reaction? Percent yield = actual yield / theoretical yield × 100 percent.

11.3/13.0 x 100% = 86.9%

- When lead (II) nitrate reacts with sodium iodide, sodium nitrate and lead (II) iodide are formed.
 - a. Balance the following equation:

 $Pb(NO_3)_{2 \text{ (sq)}} + NaI_{\text{ (sq)}} \rightarrow PbI_{2 \text{ (s)}} + NaNO_{3 \text{ (sq)}}$ $Pb(NO_3)_{2 \text{ (sq)}} + 2 NaI_{\text{ (sq)}} \rightarrow PbI_{2 \text{ (s)}} + 2 NaNO_{3 \text{ (sq)}}$

b. If I start with 25.0 grams of lead (II) nitrate and 15.0 grams of sodium iodide, how many grams of sodium nitrate can be formed?

 $25.0 \text{ g Pb(NO₃)}_2 \times 1 \text{ mol Pb(NO₃)}_2 \times 2 \text{ mol NaNO₃} \times 84.997 \text{g NaNO₃} = 12.8 \text{ g NaNO₃} = 331.214 \text{ g Pb(NO₃)}_2 + 1 \text{ mol Pb(NO₃)}_2 + 1 \text{ mol NaNO₃}_3$

15.0 g Nal x 1 mol Nal x 2 mol NaNO₃ x 84.997g NaNO₃ = 8.51 g NaNO₃ 149.89g Nal 2 mol Nal 1 mol NaNO₃

8.51 g NaNO3 can be made

Limiting Reagent And Percent Yield Key

L Darling-Hammond

Limiting Reagent And Percent Yield Key:

Limiting Reagent And Percent Yield Key: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have charmed audiences this year. The Must-Read: Colleen Hoovers "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best: Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Limiting Reagent And Percent Yield Key: Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a guiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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