


How Substances Dissolve

KEY IDEAS

As you read this section, keep these questions in mind:


- Why is water called the universal solvent?
- How do substances dissolve?

Why Do Substances Dissolve in Certain Solvents?

Water is often called a universal solvent because many substances dissolve in it. However, no one substance can dissolve every solute. A general rule in chemistry is that “like dissolves like.” This rule means that a solvent will dissolve substances that have similar molecular structures. 

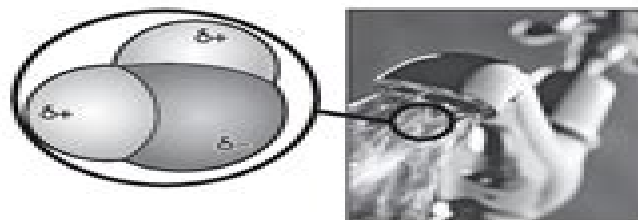
A solvent must be able to attract solute particles and pull them away from one another. Thus, solvent particles must attract solute particles more strongly than the solute particles attract one another.

POLAR MOLECULES

Recall that atoms bonded together do not all share electrons equally. For example, a water molecule is made up of two hydrogen atoms bonded to one oxygen atom. The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom. This gives each hydrogen atom a partial positive charge (δ^+) and the oxygen atom a partial negative charge (δ^-). 

Molecules that have partially charged positive and negative areas are **polar**. Because water molecules are polar, water can dissolve many other polar substances. Water can also dissolve ionic compounds such as table salt.

Water is a polar molecule because the oxygen atom strongly attracts electrons. Because the electrons spend more time near the oxygen atom, the oxygen atom has a partial negative charge.



READING TOOLBOX

Summarize Read each page of the section silently to yourself. With a partner, take turns summarizing each page.

READING CHECK

1. Identify Why is water called a universal solvent?

because
many substances dissolve
in it

READING CHECK

2. Explain Why does a water molecule have partial charges?

The oxygen atom attracts
electrons more strongly
than the hydrogen atoms
do. Thus, the electrons
spend more time near the
oxygen atom.

How Substances Dissolve Answer Key

J Rink



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Science Higher Brian Arnold,Hannah Kingston,Emma Poole,2006 This volume contains everything students need to know for Key Stage 3 higher science The text is laid out in sound bite boxes to aid recollection with clearly labelled diagrams to add visual clarity and further demonstrate the subject matter **Science I Essential Interactions** ,2000-10

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Scientifica David Sang, Peter Ellis, Derek McMonagle, 2004 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

Explaining Primary Science Paul Chambers, Nicholas Souter, 2017-03-25 Successful science teaching in primary schools requires a careful understanding of key scientific knowledge This book covers all the major areas of science relevant for beginning primary school teachers explaining key concepts from the ground up helping trainees develop into confident science educators Classroom activities and Videos of useful science experiments and demonstrations for the primary classroom are integrated into each chapter to translate concepts into teaching practice Chapter content is linked to the National Curriculum in England and the Curriculum for Excellence demonstrating how you could relate understanding to the relevant curriculum taught in schools

Chemistry for AQA. Ann Fullick, Patrick Fullick, 2001 This resource has separate books for biology chemistry and physics Each book is accompanied by a teacher's resource pack on customizable CD ROM or as a printed pack The series is designed to work in conjunction with the Coordinated Science for AQA series so that coordinated and separate science can be taught alongside each other

Contemporary Advances in Science and Technology Vol-1 Raman Singh, Bhawna Pareek, 2021-09-30 The present volume of Contemporary Advances in Science Technology is focused on advances in chemical and Biological Sciences These includes Pesticides Medicinal Plants Utilized in Marketed anti Arthritic Formulations Inorganic Ion Exchangers Organic Farming Ethical and Practical Implications of Biological Patents Nanomaterials and its Synthesis and Characterization 4

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Matter, 2004 Everything in the universe regardless of its size shape color or physical state is made up of matter

The Really Useful Science Book Steve Farrow, 1999 This book has been designed to support and extend both teachers and students own knowledge and understanding of science using accessible language to explain ideas and concepts It will be of particular interest to those who are non specialists

Building Foundations of Scientific Understanding Bernard J. Nebel Ph.D., 2014-10-09 Building Foundations of Scientific Understanding BFSU BFSU is for teachers homeschoolers and other educators to deliver a first rate science education to K 8 students and older beginning science learners Vol I here is for grades K 2 and older

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Environmental Science and Technology Frank R. Spellman, 2017-09-15 The third edition of Environmental Science and Technology Concepts and Applications is the first update since 2006 Designed for the student and the professional this newly updated reference uses scientific laws principles models and concepts to provide a basic foundation for understanding and evaluating the impact that chemicals and technology have on the environment Building upon the success of previous editions this fully revised edition has been expanded and completely updated with significant changes in the treatment of all subject areas Extensive energy parameters have been added to the text along with a thorough discussion of non renewable and renewable energy supplies and their potential impact on the environment In addition thought provoking questions have been added at the end of each chapter Finally pictorial presentation has been enhanced by the addition of numerous photographs Organization and Content Environmental Science and Technology Concepts and Applications is divided into five parts and twenty five chapters and organized to provide an even and logical flow of concepts It provides the student with a clear and thoughtful picture of this complex field Part I provides the foundation for the underlying theme of this book the connections between environmental science and technology Part II develops the air quality principles basic to an understanding of air quality Part III focuses on water quality and the characteristics of water and water bodies water sciences water pollution and water wastewater treatment Part IV deals with soil science and emphasizes soil as a natural resource highlighting the many interactions between soil and other components of the ecosystem Part V is devoted to showing how decisions regarding handling solid and hazardous waste have or can have profound impact on the environment and the three media discussed in this text air water and soil Finally the epilogue looks at the state of the environment past present and future The emphasis in this brief unit is on mitigating present and future environmental concerns by incorporating technology into the remediation process not by blaming technology for the problem

Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry

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SuperSimple Chemistry DK, 2020-06-09 A fantastic aid for coursework homework and studying for tests this comprehensive guide covers Next Generation Science Standards for grades 6 10 and will have you ready for tests and exams in no time Each topic is fully illustrated to support the information make the facts crystal clear and bring the science to life A large central image explains the idea visually and each topic is summed up on a single page helping children to quickly get up to speed and really understand how chemistry works Information boxes explain the theory with the help of simple graphics and for further studying a handy Key Facts box provides a simple summary you can check back on later With clear concise coverage of all the core topics SuperSimple Chemistry is the perfect accessible guide to chemistry for children supporting classwork and making studying for exams the easiest it s ever been

Chemical Interactions ,2005

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