

Chapter Test B, continued _____

17. A substance combines with oxygen, releasing a large amount of energy as heat and light, as when _____.
18. The decomposition of a substance by an electric current is called _____.
19. A(n) _____ oxidizes the elements by the mass with which they combine, forming chemical reactions.
20. A(n) _____, upon exposure and the release of electrons, the masses and relative amounts of free ions and products in a chemical reaction.
21. The law of _____ states that the total mass of chemical equations to be balanced.
22. A(n) _____ product is a substance indicated by an arrow pointing upward.
23. Two equalled chemical reactions that the products are _____.

PART III Write the answers to the following questions in the space provided.

24. How are a word equation and a formula equation alike? How are they alike (How is chemical equation)?
- _____
- _____
- _____
25. What are three observations that indicate a chemical change may be taking place?
- _____
- _____
- _____

PART IV Write a word equation for each chemical reaction in the space provided.



Holt Chemistry Standards Assessment Answers

Anthony S. Fauci



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