

Chapter Test Chapter 10: Matter and Energy

Matter and Energy

Write each answer in the space provided on the paper at the back of this notebook. Show all work for problems that have multiple steps.

1. ☐ 1. The mass of a proton is 1.67×10^{-27} kg. What is the mass of a neutron in grams?
 a. 1.67×10^{-27} g
 b. 1.67×10^{-24} g
 c. 1.67×10^{-21} g
 d. 1.67×10^{-18} g
2. ☐ 2. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the kinetic energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
3. ☐ 3. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the potential energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
4. ☐ 4. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the total energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
5. ☐ 5. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the internal energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
6. ☐ 6. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the enthalpy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
7. ☐ 7. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the Gibbs free energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
8. ☐ 8. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the Helmholtz free energy of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
9. ☐ 9. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the chemical potential of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J
10. ☐ 10. A sample of a substance is heated. The temperature of the sample increases from 20°C to 40°C . The mass of the sample is 100 g. What is the change in the chemical potential of the sample?
 a. 100 J
 b. 200 J
 c. 400 J
 d. 800 J

Holt Chemistry Solutions Test Answer Key

Tobias Bleicker

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investment in digital teaching and learning projects finally the book provides an explanation of the future capacity and sustainability of digital teaching and learning in k12 schools teaching mathematics in the visible learning

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