

**Gas Stoichiometry Practice 1.0 - Answer Key****Objective:** - Perform stoichiometry calculations of gases including those at STP using molar volume**Directions:** Solve each gas stoichiometry problem. Show all work. Include units.

1. Calculate the volume of hydrogen gas that will react with exactly 4.6 L of oxygen gas using the equation below. Assume there is no change in temperature or pressure.



$$4.6 \text{ L O}_2 \left( \frac{2 \text{ L H}_2}{1 \text{ L O}_2} \right) = \boxed{9.2 \text{ L H}_2}$$

2. What volume of  $\text{CO}_2$  gas will be produced from 17.5 L of oxygen gas reacting with excess ethane?



$$17.5 \text{ L O}_2 \left( \frac{4 \text{ L CO}_2}{7 \text{ L O}_2} \right) = \boxed{10.0 \text{ L CO}_2}$$

3. What volume of hydrogen at STP can be produced from the reaction of 36.5 g of zinc with hydrochloric acid?



$$36.5 \text{ g Zn} \left( \frac{1 \text{ mol Zn}}{65.38 \text{ g Zn}} \right) \left( \frac{1 \text{ mol H}_2}{1 \text{ mol Zn}} \right) \left( \frac{22.4 \text{ L H}_2}{1 \text{ mol H}_2} \right) = \boxed{12.5 \text{ L H}_2}$$

4. How many grams of NaCl can be produced by the reaction of 12.6 L of chlorine at STP with an excess of sodium?



$$12.6 \text{ L Cl}_2 \left( \frac{1 \text{ mol Cl}_2}{22.4 \text{ L Cl}_2} \right) \left( \frac{2 \text{ mol NaCl}}{1 \text{ mol Cl}_2} \right) \left( \frac{58.44 \text{ g NaCl}}{1 \text{ mol NaCl}} \right) = \boxed{65.7 \text{ g NaCl}}$$

5. In order to produce 50.0 liters of ammonia gas ( $\text{NH}_3$ ) at STP, how many grams of both hydrogen and nitrogen would you need to react?



$$50.0 \text{ L NH}_3 \left( \frac{1 \text{ mol NH}_3}{22.4 \text{ L NH}_3} \right) \left( \frac{1 \text{ mol N}_2}{2 \text{ mol NH}_3} \right) \left( \frac{28.02 \text{ g N}_2}{1 \text{ mol N}_2} \right) = \boxed{21.3 \text{ g N}_2}$$

$$50.0 \text{ L NH}_3 \left( \frac{1 \text{ mol NH}_3}{22.4 \text{ L NH}_3} \right) \left( \frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3} \right) \left( \frac{2.02 \text{ g H}_2}{1 \text{ mol H}_2} \right) = \boxed{6.75 \text{ g H}_2}$$

# Gas Stoichiometry Practice With Answers

**Kaplan Test Prep**



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**Dr. Salm's Chemistry Problem Drill Book** Robert J. Salm, Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests **The Practice of**

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