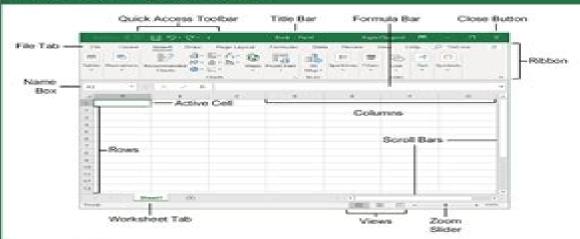




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Copy	Ctrl - C
Payle	Otri - V
Undo	Ctrl - Z
Redo	Ctrl - Y
First	Ctrl + F
Replace	Cos - H
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Clear cell contents	Delete

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Italics	Otri - I
Underline	Ctrl + U
Open Format Cells dialog box	Ctrl - Shift
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Greg Harvey

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the research instrument consisted of a questionnaire to measure students attitudes towards stem and multiple choice tests to assess students cognitive learning outcomes

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sugiyono 2011 90 minimalized sample of the research are 30 people in this research the researcher used the last option because amount of the the researcher use this instrument to know how far students understanding about skimming skill by giving answer in the questionnare based on the guidance for skimming mc whorter 1992 81

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showed levels of research instrument validity arikunto 2010 211 degradation of accuracy between data which was happen on research object and capacity which can be reported by researcher sugiyono 2014 267 high and low validity of instrument showed how far data were collected didn t diverge from validity itself

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sugiyono 2012 102 state that research instrument is the tools to measure nature phenomenon and social that is observed research instrument is essentially used to get information or data about the object taken from subject of the research many kinds of instrument which can be used in a research such as test questionnaires interview

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3 3 1 technique and instrument according to sugiyono 2012 148 the research instrument is a tool used to measure the natural and social phenomena observed the instruments for collecting the data is interview guide observation check list and documents analysis

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the purpose of this study was to produce student worksheet lembar kerja peserta didik lkpd and pisa based test instruments that were valid practical effective and able to improve students mathematical representation abilities this type of research was the tessmer model development research

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the

titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as ca 3 po 4 2 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore k ca2 3 po 4 2 ca3 po 4 2 ca3 po 4 2 ca3 po 4 2 k ksp ca2 3 po 3 4 2

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value to the theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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