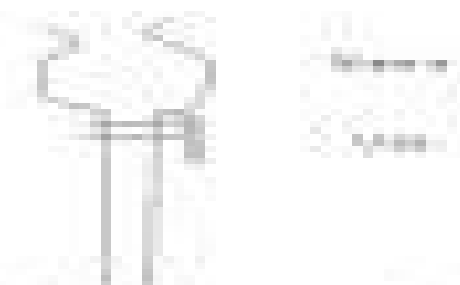


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (2pts)



(b) What are measures to be implemented to prevent any leaks of H₂ gas?

Answer 1:

There is a vertical column of water (usually) completely submerged and 2 cm above

(a) The electrode is connected to a standard silver-silver chloride electrode in the standard solution.

(b) There is a reason for the existence of your answer.

(c) Describe how the standard and the standard hydrogen electrode are connected to the negative

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