

|        |                                |          |       |
|--------|--------------------------------|----------|-------|
| Page 2 | Mark Scheme: Teachers' version | Syllabus | Paper |
|        | IGCSE – October/November 2010  | 0620     | 22    |

- 1 (a) magnesium oxide / MgO [1]
- (b) nitrogen dioxide / NO<sub>2</sub>; [1]  
ALLOW nitrogen oxide  
sulfur dioxide / SO<sub>2</sub> [1]  
ALLOW sulfur oxide [1]
- (c) carbon dioxide / CO<sub>2</sub>; [1]  
water / H<sub>2</sub>O [1]
- (d) water / H<sub>2</sub>O [1]
- (e) carbon dioxide / CO<sub>2</sub> [1]
- [Total: 7]**
- 2 (a) (i) substance containing two (or more) different atoms / elements joined / combined / bonded [1]  
BOTH idea of different atoms / elements and bonded needed for 1 mark
- (ii) (compound) B; [1]  
it is an ionic giant structure / it is ionic [1]  
ALLOW it contains ions
- (iii) C [1]
- (b) (i) 1st box ticked (conducts when molten) [1]
- (ii) add (aqueous) silver nitrate; [1]  
(light) yellow precipitate (BOTH yellow and precipitate required) [1]  
2nd mark dependent on correct reagent  
NOT cream precipitate  
ALLOW lead nitrate (1) yellow precipitate (1)
- (c) it is an oxide of a non-metal / iodine is a non-metal [1]
- [Total: 8]**

# Marking Scheme 0620 11 M J 14

**Puja Mehta**



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## **Table of Contents Marking Scheme 0620 11 M J 14**

1. Understanding the eBook Marking Scheme 0620 11 M J 14
  - The Rise of Digital Reading Marking Scheme 0620 11 M J 14
  - Advantages of eBooks Over Traditional Books
2. Identifying Marking Scheme 0620 11 M J 14
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Marking Scheme 0620 11 M J 14
  - User-Friendly Interface
4. Exploring eBook Recommendations from Marking Scheme 0620 11 M J 14
  - Personalized Recommendations
  - Marking Scheme 0620 11 M J 14 User Reviews and Ratings
  - Marking Scheme 0620 11 M J 14 and Bestseller Lists
5. Accessing Marking Scheme 0620 11 M J 14 Free and Paid eBooks

- Marking Scheme 0620 11 M J 14 Public Domain eBooks
- Marking Scheme 0620 11 M J 14 eBook Subscription Services
- Marking Scheme 0620 11 M J 14 Budget-Friendly Options
- 6. Navigating Marking Scheme 0620 11 M J 14 eBook Formats
  - ePub, PDF, MOBI, and More
  - Marking Scheme 0620 11 M J 14 Compatibility with Devices
  - Marking Scheme 0620 11 M J 14 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Marking Scheme 0620 11 M J 14
  - Highlighting and Note-Taking Marking Scheme 0620 11 M J 14
  - Interactive Elements Marking Scheme 0620 11 M J 14
- 8. Staying Engaged with Marking Scheme 0620 11 M J 14
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Marking Scheme 0620 11 M J 14
- 9. Balancing eBooks and Physical Books Marking Scheme 0620 11 M J 14
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Marking Scheme 0620 11 M J 14
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Marking Scheme 0620 11 M J 14
  - Setting Reading Goals Marking Scheme 0620 11 M J 14
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Marking Scheme 0620 11 M J 14
  - Fact-Checking eBook Content of Marking Scheme 0620 11 M J 14
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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