



# Basic Chemistry



According to the author, a compound is composed of two or more elements chemically united. (Underlined) All atoms differently sized, yet identical properties like, mass, size, chemical nature, hydrogen, oxygen, and nitrogen making molecules of air being formed.

The student explains a compound is made of two or more different elements, combined chemically, forming a substance that has the properties of none of the original elements. (Underlined) The chemical composition of a compound is determined experimentally by the mass composition. (Underlined) Oxygen, hydrogen, nitrogen, carbon, and sulfur are the most important elements in forming compounds. (Underlined) They are called main, life, or organic.

Compounds are made of two or more different elements, and the chemical and physical properties of the compound are different from the properties of the elements. (Underlined) The chemical composition of a compound is determined by the mass composition. (Underlined) They are called main, life, or organic.

## CONCEPTS OF MATTER AND ENERGY

1. Matter defines the region with a following characteristic and mass. (Underlined) Matter is defined as anything that has mass and occupies space.

(Underlined) The matter is defined by the following characteristics:

- |                         |                                |
|-------------------------|--------------------------------|
| a. is continuous matter | b. matter is continuous matter |
| c. is continuous matter | d. matter is continuous matter |

(Underlined) Matter is defined by the following characteristics:

- |                                |                                |
|--------------------------------|--------------------------------|
| a. matter is continuous matter | b. matter is continuous matter |
| c. matter is continuous matter | d. matter is continuous matter |

(Underlined) Matter is defined by the following characteristics:

- |                                |                                |
|--------------------------------|--------------------------------|
| a. matter is continuous matter | b. matter is continuous matter |
| c. matter is continuous matter | d. matter is continuous matter |

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