Limiting Reactants Practice

1. 2 NO + __O₂ → 2 NO₂

a) What is the limiting reactant if 0.866 mol of NO is mixed with 0.503 mol of O2 to create NO2?

 b) Calculate the number of moles of NO₂ produced. (Answer should already be solved from the previous questions)

0.866 mol NO

a) If 7.40 g of O2 reacts with 0.670 g of NO, which compound will be the limiting reactant?

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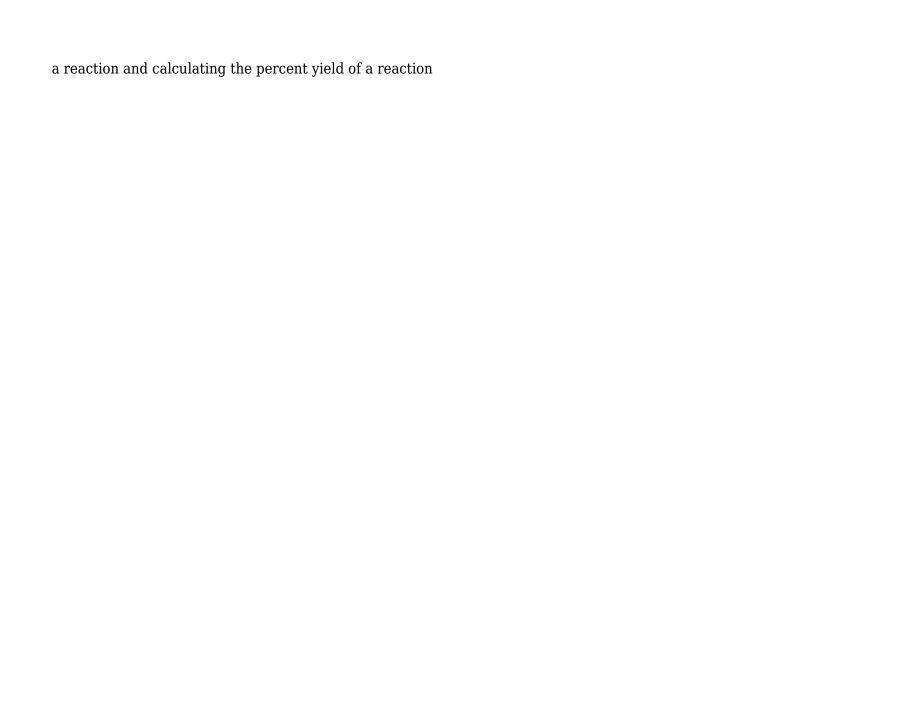
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