

## Limiting Reagent Worksheet

*Using your knowledge of stoichiometry and limiting reagents, answer the following questions:*

- 1) Write the balanced equation for the reaction of lead (II) nitrate with sodium iodide to form sodium nitrate and lead (II) iodide:
  
  
  
  
  
  
  
  
  
  
- 2) If I start with 25.0 grams of lead (II) nitrate and 15.0 grams of sodium iodide, how many grams of sodium nitrate can be formed?
  
  
  
  
  
  
  
  
  
  
- 3) What is the limiting reagent in the reaction described in problem 2?
  
  
  
  
  
  
  
  
  
  
- 4) How much of the nonlimiting reagent will be left over from the reaction in problem #2?

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