



Gizmos

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Answer these questions BEFORE using the Gizmo.)

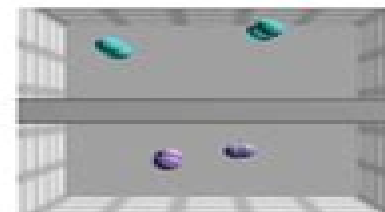
Imagine you and your friends are making hot dogs. A complete hot dog consists of a frankfurter and a bun. At the store, you buy four packages of eight frankfurters and three bags of 10 buns.

1. How many hot dogs can you make? 30
2. Which ingredient limited the number of hot dogs you could make? The buns
3. Which ingredient will you have leftovers of? 2 hot dogs left over


Gizmo Warm-up

Just as ingredients can be put together to make new food items, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants Gizmo™*, you can determine which reactant is limiting in various scenarios.



To begin, make sure $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of O_2 molecules and H_2 molecules to 2.
 - A. How many hydrogen molecules (H_2) are there? 2
Hydrogen atoms? 4
 - B. How many oxygen molecules (O_2) are there? 2
Oxygen atoms? 4
2. How many H_2O molecules do you think will form when these 4 molecules react? 2 molecules
3. Click Play (). How many H_2O molecules actually formed? 2 molecules



Limiting Reactant Gizmo Teachers Guide

Jiyuan Zhang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Limiting Reactant Gizmo Teachers Guide

1. Understanding the eBook Limiting Reactant Gizmo Teachers Guide
 - The Rise of Digital Reading Limiting Reactant Gizmo Teachers Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Limiting Reactant Gizmo Teachers Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Limiting Reactant Gizmo Teachers Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Limiting Reactant Gizmo Teachers Guide

- Personalized Recommendations
- Limiting Reactant Gizmo Teachers Guide User Reviews and Ratings
- Limiting Reactant Gizmo Teachers Guide and Bestseller Lists
- 5. Accessing Limiting Reactant Gizmo Teachers Guide Free and Paid eBooks
 - Limiting Reactant Gizmo Teachers Guide Public Domain eBooks
 - Limiting Reactant Gizmo Teachers Guide eBook Subscription Services
 - Limiting Reactant Gizmo Teachers Guide Budget-Friendly Options
- 6. Navigating Limiting Reactant Gizmo Teachers Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Limiting Reactant Gizmo Teachers Guide Compatibility with Devices
 - Limiting Reactant Gizmo Teachers Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Limiting Reactant Gizmo Teachers Guide
 - Highlighting and Note-Taking Limiting Reactant Gizmo Teachers Guide
 - Interactive Elements Limiting Reactant Gizmo Teachers Guide
- 8. Staying Engaged with Limiting Reactant Gizmo Teachers Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Limiting Reactant Gizmo Teachers Guide
- 9. Balancing eBooks and Physical Books Limiting Reactant Gizmo Teachers Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Limiting Reactant Gizmo Teachers Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Limiting Reactant Gizmo Teachers Guide
 - Setting Reading Goals Limiting Reactant Gizmo Teachers Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Limiting Reactant Gizmo Teachers Guide

- Fact-Checking eBook Content of Limiting Reactant Gizmo Teachers Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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