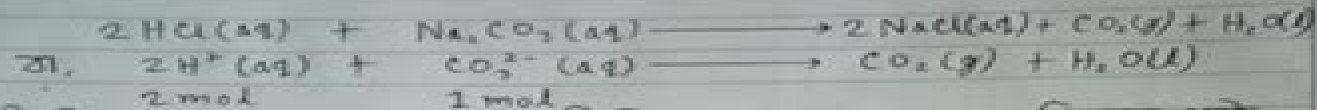


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকটগ্রস্ত নমুনা উত্তোলন (HCl উত্তোলন)	Expt No.	01
চলমান নিন্দিত	Date :	
মূলনীতি:		

Na_2CO_3 ও HCl এর প্রকৃষ্ট বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃষ্ট বিক্রিয়া কোষ প্রকৃষ্ট বিক্রিয়া করে, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যারি) থেকে দেয়া।

প্রয়োজনীয় সামগ্র্যনিক বস্তু:

- ১) 0.1M Na_2CO_3 দ্রব
- ২) 0.1M HCl দ্রব
- ৩) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- ১) কনিক্যাল ফ্লাস্ক
- ২) ক্যারি
- ৩) পিপেট
- ৪) ক্যাম্পাস মেসার

কার্যের ধারা:

- ১) প্রস্তুত করা 0.1M HCl দ্রব (প্রকৃষ্ট দ্রব) অংশ নমুনা HCl দ্বারা ক্যারিটি বর্ণ করে এবং মেসারের মাধ্যমে ক্যাম্পাস দ্বারা পরিমাপ করে আটকিয়ে নিই।
- ২) প্রস্তুত করা 0.1M Na_2CO_3 দ্রব (প্রকৃষ্ট দ্রব) এর আণুবিন্যাস ফ্রাক্ট থেকে পিপেট দ্বারা 10mL দ্রব কনিক্যাল ফ্লাস্কে নিই, কনিক্যাল ফ্লাস্কে দ্রব 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, তখন দ্রবের বর্ণ হলুদ হয়।
- ৩) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখি, ক্যারি HCl এমি দ্রবের মতি নিয়ে ঘোঁটা করে '১ম পাই' - এ ফেরে করে, চিহ্নমতে রাস মাতে ক্যারিটের মেন লক ঘুরিয়ে ক্যারিটের HCl দ্রব ঘোঁটা ঘোঁটা কনিক্যাল ফ্লাস্কে যোগ করে

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R Barnett



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Table of Contents Hsc Chemistry 2nd Paper Practical 23

1. Understanding the eBook Hsc Chemistry 2nd Paper Practical 23
 - The Rise of Digital Reading Hsc Chemistry 2nd Paper Practical 23
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Chemistry 2nd Paper Practical 23
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Chemistry 2nd Paper Practical 23
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Chemistry 2nd Paper Practical 23
 - Personalized Recommendations

- Hsc Chemistry 2nd Paper Practical 23 User Reviews and Ratings
- Hsc Chemistry 2nd Paper Practical 23 and Bestseller Lists
- 5. Accessing Hsc Chemistry 2nd Paper Practical 23 Free and Paid eBooks
 - Hsc Chemistry 2nd Paper Practical 23 Public Domain eBooks
 - Hsc Chemistry 2nd Paper Practical 23 eBook Subscription Services
 - Hsc Chemistry 2nd Paper Practical 23 Budget-Friendly Options
- 6. Navigating Hsc Chemistry 2nd Paper Practical 23 eBook Formats
 - ePub, PDF, MOBI, and More
 - Hsc Chemistry 2nd Paper Practical 23 Compatibility with Devices
 - Hsc Chemistry 2nd Paper Practical 23 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hsc Chemistry 2nd Paper Practical 23
 - Highlighting and Note-Taking Hsc Chemistry 2nd Paper Practical 23
 - Interactive Elements Hsc Chemistry 2nd Paper Practical 23
- 8. Staying Engaged with Hsc Chemistry 2nd Paper Practical 23
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Chemistry 2nd Paper Practical 23
- 9. Balancing eBooks and Physical Books Hsc Chemistry 2nd Paper Practical 23
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Chemistry 2nd Paper Practical 23
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Chemistry 2nd Paper Practical 23
 - Setting Reading Goals Hsc Chemistry 2nd Paper Practical 23
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Chemistry 2nd Paper Practical 23
 - Fact-Checking eBook Content of Hsc Chemistry 2nd Paper Practical 23

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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