

# HOLT MODERN CHEMISTRY -- CHAPTER 1 REVIEW

*The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test. \*\*\*\*\* Test items will be recall, examples, and/or application of this content. \*\*\*\*\**

## OUTCOMES

- 1.1: Distinguish between types of research – basic, applied, and technological development. (F & PK)
- 1.2: Distinguish between the physical and chemical properties of matter. (F & PK)
- 1.2: Distinguish among homogeneous and heterogeneous mixtures, elements, and compounds. (F & PK)
- 1.3: List properties of metals and non-metals. (T & R)
- 2.3: Determine significant figures. (F & PK)

## 1.1: CHEMISTRY IS A PHYSICAL SCIENCE

- **Vocabulary**
  - **chemistry** -- the scientific study of the composition, structure, and properties of matter and the changes that matter undergoes
  - **chemical** -- any substance with a defined composition
- **Chapter Highlights**
  - Chemistry is the study of the composition, structure, and properties of matter and the changes that matter undergoes.
  - A chemical is any substance that has a definite composition or is used or produced in a chemical process.
  - Basic research is carried out for the sake of increasing knowledge. Applied research is carried out to solve practical problems. Technological development involves the use of existing knowledge to make life easier or more convenient.

## 1.2: MATTER AND ITS PROPERTIES

- **Vocabulary**
  - **matter** -- anything that has mass and takes up space
  - **mass** -- a measure of the amount of matter in an object
  - **element** -- a substance that cannot be separated or broken down into simpler substances by chemical means; all atoms of an element have the same atomic number
  - **atom** -- the smallest unit of an element that maintains the chemical properties of that element
  - **compound** -- a substance made up of atoms of two or more different elements joined by chemical bonds
  - **extensive property** -- a property that depends on the extent or size of a system
  - **intensive property** -- a property that does not depend on the amount of matter present, such as pressure, temperature, or density
  - **physical property** -- a characteristic of a substance that does not involve a chemical change, such as density, color, or hardness
  - **physical change** -- a change of matter from one form to another without a change in chemical properties
  - **change of state** -- the change of a substance from one physical state to another
  - **solid** -- the state of matter in which the volume and shape of a substance are fixed
  - **liquid** -- the state of matter that has a definite volume but not a definite shape
  - **gas** -- a form of matter that does not have a definite volume or shape
  - **plasma** -- in physical science, a state of matter that starts as a gas and then becomes ionized; it consists of freemoving ions and electrons and it takes on an electric charge
  - **chemical property** -- a property of matter that describes a substance's ability to participate in chemical reactions
  - **chemical change** -- a change that occurs when one or more substances change into entirely new substances with different properties
  - **chemical reaction** -- the process by which one or more substances change to produce one or more different substances
  - **reactant** -- a substance or molecule that participates in a chemical reaction
  - **product** -- a substance that forms in a chemical reaction
  - **mixture** -- a combination of two or more substances that are not chemically combined

# Holt Chemistry Chapter Review

**Sabine Zange**

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**The School Review**, 1914      *Naval Research Reviews*, 1996      The Review of Reviews, 1895      **Holt General Science** William L. Ramsey, 1988      *The St. Louis Medical Review*, 1900      **Bulletin** Yale University, 1928      *Books in Print Supplement*, 2002      **Allergy and Allergic Diseases** A. Barry Kay, Jean Bousquet, Patrick G. Holt, Allen P. Kaplan, 2009-01-26 Reserve your copy now This two volume book is an outstanding reference source on all aspects of allergy and allergic diseases Covering virtually every allergic condition from the immunological and molecular basis of the allergic response to future trends in allergic disease prevention this new international editorial team A B Kay Jean Bousquet Pat Holt and Allen Kaplan have completely revised and updated the text from both a scientific and clinical perspective References will continue to be added to the text until it goes to press making this the most up to date book available in the field This second edition consists of more than 1 800 pages contained within 98 chapters The price includes a fully searchable companion CD ROM with the complete text and over 300 images from the book in full colour      **The Prokaryotes** M.P. Starr, H. Stolp, H.G. Trüper, A. Balows, H.G. Schlegel, 2013-11-11 The purpose of this brief Foreword is to make you the reader hungry for the scientific feast that follows These two volumes on the prokaryotes offer a truly unique scientific menu a comprehensive assembly of articles exhibiting the biochemical depth and remarkable physiological and morphological diversity of prokaryote life The size of the volumes might initially discourage the unprepared mind from being attracted to the study of prokaryote life for this landmark assemblage thoroughly documents the wealth of present knowledge But in confronting the reader with the state of the art the Handbook also defines where new work needs to be done on well studied bacteria as well as on unusual or poorly studied organisms There are basically two ways of doing research with microbes A classical approach is first to define the phenomenon to be studied and then to select the organism accordingly Another way is to choose a specific

organism and go where it leads The pursuit of an unusual microbe brings out the latent hunter in all of us The intellectual challenges of the chase frequently test our ingenuity to the limit Sometimes the quarry repeatedly escapes but the final capture is indeed a wonderful experience For many of us these simple rewards are sufficiently gratifying so that we have chosen to spend our scientific lives studying these unusual creatures

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