Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry Section 1: What is Chemistry? Objectives:

- Describe ways in which chemistry is a part of your daily life.
- Describe the characteristics of three common states of matter.
- Describe physical and chemical changes, and give examples of each.
- Identify the reactants and products in a chemical reaction.
- List four observations that suggest a chemical change has occurred.

A _____ is any substance that has a definite composition. A _____ is the process by which one or more substances change to produce one or more new substances. The _____ of ____ are the physical forms of matter which are solid, liquid, gas, and plasma.

- have a fixed volume and shape that result from the way their particles are arranged.
- have a fixed volume but not a fixed shape.
 have neither fixed volume nor shape.

Water molecule, H₂O Water molecule, H₂O Water molecule, H₂O

Physical changes are changes in which the

of a substance doesn't change.

Changes of state are physical changes.

Chemical changes occur when the ______ of substances change and new substances form.

Evidence of a chemical change:



without accepta, secol, or vinegast, and soldinam hydrogen carbonate, or baking soda, are mircol, the solution bubbles as carbon dionale forms.



b When solitions of soliton solitate and endinness natrate are mixed, cadmins solitate, a solid proceputate, forms.



 When aluminum nearly with iron(EH) inside in the clay post, emergy is released in heart and light.



d When phenolphthalein is added to announce dissolved in water, a color change from coloritos to pink occurs.

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United States. National Bureau of Standards

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Calendar - McGill University McGill University,1904 Miscellaneous Publication - National Bureau of
Standards United States. National Bureau of Standards,1934 Publications of the National Bureau of Standards United States. National Bureau of Standards,1960

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Modern Alchemy Glenn Theodore Seaborg,1994 During his distinguished career spanning more than 50 years Nobel laureate Chemistry Glenn T Seaborg published over 500 works This volume puts together about 100 of his selected papers The papers are divided into five categories Category I consists of papers which detail the discovery of 10 transuranium elements and numerous heavy isotopes of special importance Category II papers describe the discovery of a number of isotopes which became the workhorses of nuclear medicine or found other applications Papers in Category III describe how the chemical properties of transuranium elements were originally determined how chemistry is applied in nuclear sciences and other chemical investigations including early work done with the great chemist G N Lewis Papers in Category IV cover radioactive decay chains and nuclear systematics Lastly papers in Category V illustrate how the powerful methods of chemistry are used to explain nuclear reactions in low intermediate and high energy nuclear physics

Supplementary List of Publications of the National Bureau of Standards, 1960 **Journal of the American Medical Association** American Medical Association, 1917 *The Engineer*, 1878 **Resources in Education**, 1988 *Chemistry*Steven S. Zumdahl, 1989 <u>Bulletin</u> United States. Bureau of Chemistry, 1891 *City and State* Herbert Welsh, 1899

The Electrician ,1916 The Electrical Journal ,1916 The Examiner ,1856 Clinical Cardiac Pacing,

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