

For each compound, write  
**IONIC** or **COVALENT**:

1

- a)  $\text{MgCl}_2$
- b)  $\text{CCl}_4$
- c)  $\text{Na}_2\text{CO}_3$
- d)  $\text{NH}_3$



Write the name of the  
following **covalent** compounds:

2

- a)  $\text{N}_2\text{O}_3$
- b)  $\text{CBr}_4$
- c)  $\text{P}_2\text{O}_5$



Write the number of  
Valence electrons for each  
element:

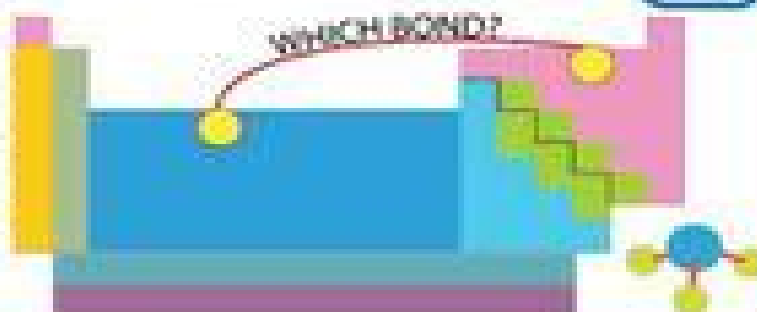
3

- a) **Phosphorus**
- b) Helium
- c) Calcium
- d) Chlorine



**Ionic or Covalent?**

4



Which Electron Dot Diagram  
is **NOT** correct? **EXPLAIN**.

5



Write the **chemical formula**  
of each compound:

6

- a) Calcium oxide
- b) Iron (II) chloride
- c) Dihydrogen dioxide



# Holt Chemistry Bonds Compounds Concept Review Answers

**Ying Liu**



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**Holt Chemistry** Holt Rinehart & Winston, 2003-01-24      **Holt Chemistry** Ralph Thomas Myers, 2004      *Modern Chemistry* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001      **Laboratory Practice**, 1979      Current Index to Journals in Education, 1978      **Choice**, 1966      **British Books in Print**, 1968      *Books in Print*, 1961      *Technical Books in Print*, 1974      **Readings**, 1986      **Holt Chemistry** Holt, Rinehart and Winston Staff, 1996      **The Concept of the Chemical Bond** Dieter Cremer, Z.B. Maksic, 1990-06-13 The state of the art in contemporary theoretical chemistry is presented in this 4 volume set with numerous contributions from the most highly regarded experts in their field It provides a concise introduction and critical evaluation of theoretical approaches in relation to experimental evidence

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