

## Chemistry Periodic Table Test Chapter 6 Chem IH

**Matching** Match each item with the correct statement below.

- |                      |                     |
|----------------------|---------------------|
| a. electronegativity | e. transition metal |
| b. ionization energy | f. periodic law     |
| c. electrons         | g. group            |
| d. metal             | h. period           |

- \_\_\_\_ 1. vertical column in the periodic table
- \_\_\_\_ 2. type of element characterized by the presence of electrons in the *d* orbital
- \_\_\_\_ 3. type of element characterized by losing electrons easily
- \_\_\_\_ 4. ability of an atom to attract electrons when the atom is in a compound
- \_\_\_\_ 5. energy required to remove an electron from an atom

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_ 6. What is another name for the main group elements?
- |                              |                         |
|------------------------------|-------------------------|
| a. s block elements          | c. s & p block elements |
| b. p block elements elements | d. d block elements     |
- \_\_\_\_ 7. What is another name for the transition metals?
- |                              |                              |
|------------------------------|------------------------------|
| a. noble gases               | c. p block elements elements |
| b. s block elements elements | d. d block elements          |
- \_\_\_\_ 8. Who arranged the elements according to atomic mass and used the arrangement to predict the properties of missing elements?
- |                      |                     |
|----------------------|---------------------|
| a. Henry Moseley     | c. John Dalton      |
| b. Antoine Lavoisier | d. Dmitri Mendeleev |
- \_\_\_\_ 9. Which of the following categories includes the majority of the elements?
- |               |              |
|---------------|--------------|
| a. metalloids | c. metals    |
| b. liquids    | d. nonmetals |
- \_\_\_\_ 10. In which of the following sets is the symbol of the element, the number of protons, and the number of electrons given correctly?
- |                                 |                                    |
|---------------------------------|------------------------------------|
| a. In, 49 protons, 49 electrons | c. Cs, 55 protons, 132.9 electrons |
| b. Zn, 30 protons, 60 electrons | d. F, 19 protons, 19 electrons     |
- \_\_\_\_ 11. The atomic number of an element is the total number of which particles in the nucleus?
- |             |                          |
|-------------|--------------------------|
| a. neutrons | c. electrons             |
| b. protons  | d. protons and electrons |
- \_\_\_\_ 12. What element has the electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^2$ ?
- |             |            |
|-------------|------------|
| a. nitrogen | c. silicon |
| b. selenium | d. silver  |
- \_\_\_\_ 13. Which subatomic particle plays the greatest part in determining the properties of an element?
- |             |                      |
|-------------|----------------------|
| a. proton   | c. neutron           |
| b. electron | d. none of the above |
- \_\_\_\_ 14. Which of the following elements is a transition metal?
- |           |              |
|-----------|--------------|
| a. cesium | c. tellurium |
| b. copper | d. tin       |

# Holt Chemistry The Periodic Table Chapter Test

**Clifford Lane**



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