

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Table of Contents Holt Chemistry Chapterreview Answers

1. Understanding the eBook Holt Chemistry Chapterreview Answers
 - The Rise of Digital Reading Holt Chemistry Chapterreview Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Chapterreview Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Chapterreview Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Chapterreview Answers
 - Personalized Recommendations
 - Holt Chemistry Chapterreview Answers User Reviews and Ratings
 - Holt Chemistry Chapterreview Answers and Bestseller Lists
5. Accessing Holt Chemistry Chapterreview Answers Free and Paid eBooks
 - Holt Chemistry Chapterreview Answers Public Domain eBooks
 - Holt Chemistry Chapterreview Answers eBook Subscription Services
 - Holt Chemistry Chapterreview Answers Budget-Friendly Options

6. Navigating Holt Chemistry Chapterreview Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Chapterreview Answers Compatibility with Devices
 - Holt Chemistry Chapterreview Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Chapterreview Answers
 - Highlighting and Note-Taking Holt Chemistry Chapterreview Answers
 - Interactive Elements Holt Chemistry Chapterreview Answers
8. Staying Engaged with Holt Chemistry Chapterreview Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Chapterreview Answers
9. Balancing eBooks and Physical Books Holt Chemistry Chapterreview Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Chapterreview Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Chemistry Chapterreview Answers
 - Setting Reading Goals Holt Chemistry Chapterreview Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Chemistry Chapterreview Answers
 - Fact-Checking eBook Content of Holt Chemistry Chapterreview Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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