



CHEMISTRY

A CHEAT SHEET

EQUILIBRIUM

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<p>Equilibrium Equilibrium?</p> <p>Equilibrium is a state of balance between two or more opposing forces or processes. In chemistry, it refers to a state where the concentrations of reactants and products remain constant over time.</p> <p>At equilibrium, the rate of the forward reaction is equal to the rate of the reverse reaction.</p> <p>Equilibrium Constant (K_{eq}) $K_{eq} = \frac{[Products]}{[Reactants]}$</p>	<p>Equilibrium Constant (K_{eq}) $K_{eq} = \frac{[Products]}{[Reactants]}$</p> <p>Equilibrium Constant (K_{eq}) $K_{eq} = \frac{[Products]}{[Reactants]}$</p> <p>Equilibrium Constant (K_{eq}) $K_{eq} = \frac{[Products]}{[Reactants]}$</p>	
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