

## (Solution Stoichiometry, Precipitation Reactions, Acid/Base Reactions, Multi-Concept Reaction Stoichiometry)

Solution Stoichiometry: Molarity, Dilutions and Working with Density

1. Calculate the molarity of each of the following solutions:

a. 2.026 g of  $\text{FeCl}_3$  in 0.1250 L of solution

$$\frac{2.026 \text{ g FeCl}_3}{0.1250 \text{ L}} \times \frac{1 \text{ mole FeCl}_3}{162.2 \text{ g}} = 0.09993 \text{ M}$$

b. 0.001 mg of  $\text{Cd}^{2+}$  in 0.100 L, the maximum permissible concentration of cadmium in drinking water

$$\frac{0.001 \text{ mg Cd}^{2+}}{0.100 \text{ L}} \times \frac{1 \text{ g}}{1000 \text{ mg}} \times \frac{1 \text{ mole Cd}^{2+}}{112.41 \text{ g Cd}^{2+}} = \sim 9 \times 10^{-8} \text{ M}$$

2. Convert 1b to ppm and ppb.

$$\text{ppm} = \text{Parts per million} \quad \text{i.e.} \quad \frac{\text{g}}{10^6 \text{ g}} \text{ or } \frac{\text{g}}{10^6 \text{ mL}} \text{ or } \frac{\text{mg}}{\text{L}}$$

$$\text{ppb} = \text{Parts per billion} \quad \text{i.e.} \quad \frac{\text{g}}{10^9 \text{ g}} \text{ or } \frac{\text{g}}{10^9 \text{ mL}} \text{ or } \frac{\mu\text{g}}{\text{L}}$$

$$\frac{0.001 \text{ mg Cd}^{2+}}{0.100 \text{ L}} \times \frac{1 \text{ ppm}}{\text{mg/L}} = 0.01 \text{ ppm}$$

$$\frac{0.001 \text{ mg Cd}^{2+}}{0.100 \text{ L}} \times \frac{1 \text{ g}}{1000 \text{ mg}} \times \frac{10^6 \mu\text{g}}{1 \text{ g}} \times \frac{1 \text{ ppb}}{\mu\text{g/L}} = 10 \text{ ppb}$$

# Gas Law Stoichiometry Worksheet Answers

**Sydney Young**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Gas Law Stoichiometry Worksheet Answers:

*General Chemistry Workbook* Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

*Introduction to Chemistry, Laboratory Manual* T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

**Chemistry**, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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