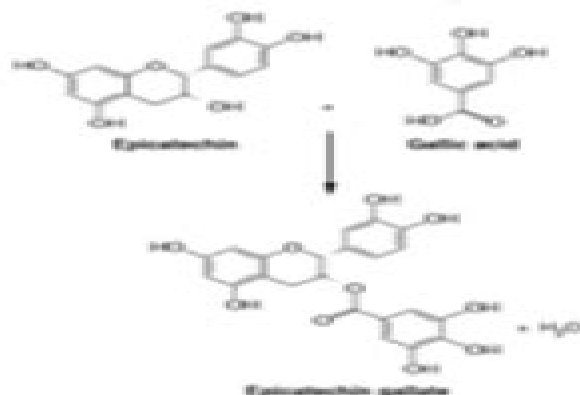


F332 Adv. Notice June 2014 : some "predicted Qs"

Selection of student-suggested Qs based on the advance notice material for F332 June 2014,in no particular order.....

Q1] TCA only needs to be present at about 2 parts per trillion for a bottle of wine to be ruined. What is this as a percentage? (1,000,000 million is a trillion)



Q2] In figure 3 (above), epicatechin and gallic acid react together to form Epicatechin gallate. On the diagram above, circle the carboxylic acid functional group on the Gallic acid molecule. This is known as a "condensation reaction". Suggest why?

Q3] Butanedione, a molecule formed during the aging of Chardonnay wines, is often suspected of being the cause of the buttery taste. Draw the skeletal formula of this molecule.

Q4] Vitamin C (ascorbic acid) is found in the grapes. The molecule contains more than one type of alcohol. On the molecule below, circle and label the different types of alcohol present.



Q5] Oct-1-en-3-ol is a molecule suspected of causing a bad taste in corked wine.



- Suggest a structural isomer of this molecule – draw its structural formula and name it.
- Suggest the reagents and conditions needed to convert oct-1-en-3-ol into oct-1-en-3-one in a chemistry lab.
- Suggest how Infra-Red spectrometry could be used to differentiate between molecules of oct-1-en-3-ol and oct-1-en-3-one.

F332 June 2014 Paper

Christian G. Meyer



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Table of Contents F332 June 2014 Paper

1. Understanding the eBook F332 June 2014 Paper
 - The Rise of Digital Reading F332 June 2014 Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying F332 June 2014 Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an F332 June 2014 Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from F332 June 2014 Paper
 - Personalized Recommendations
 - F332 June 2014 Paper User Reviews and Ratings
 - F332 June 2014 Paper and Bestseller Lists
5. Accessing F332 June 2014 Paper Free and Paid eBooks

- F332 June 2014 Paper Public Domain eBooks
- F332 June 2014 Paper eBook Subscription Services
- F332 June 2014 Paper Budget-Friendly Options
- 6. Navigating F332 June 2014 Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - F332 June 2014 Paper Compatibility with Devices
 - F332 June 2014 Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of F332 June 2014 Paper
 - Highlighting and Note-Taking F332 June 2014 Paper
 - Interactive Elements F332 June 2014 Paper
- 8. Staying Engaged with F332 June 2014 Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers F332 June 2014 Paper
- 9. Balancing eBooks and Physical Books F332 June 2014 Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection F332 June 2014 Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine F332 June 2014 Paper
 - Setting Reading Goals F332 June 2014 Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of F332 June 2014 Paper
 - Fact-Checking eBook Content of F332 June 2014 Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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