

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons)

Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

Eoc Chemistry Study Guide

David Matheson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Eoc Chemistry Study Guide:

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Pamphlet - Dept. of the Army United States. Department of the Army, 1951

Books In Print 2004-2005 Ed Bowker Staff, Staff Bowker, Ed, 2004

Newly Hired Teachers of Science Julie A. Luft, Shannon L. Dubois, 2015-12-09 Supporting newly hired science teachers has taken on an increased importance in our schools This book shares the most current information about the status of newly

hired science teachers different ways in which to support newly hired science teachers and different research approaches that can provide new information about this group of teachers Chapters in the book are written by those who study the status of beginning science teachers mentor new teachers develop induction programs and research the development of new science teachers Newly Hired Teachers of Science is for administrators who have new science teachers in their schools and districts professionals who create science teacher induction programs mentors who work closely with new science teachers educational researchers interested in studying new science teachers and even new science teachers This is a comprehensive discussion about new science teachers that will be a guiding document for years to come **Exemplary Science in Grades**

9-12 Robert Eugene Yager, 2005 Essay titles reveal the range of programs and creativity this book encompasses The book ends with a summary chapter on successes and continuing challenges in meeting the Standards visions for improving high school science **Life of the Soldier and the Airman**, 1951 *Standard Nomenclature List* United States Armed Forces Institute, 1953 **An Introduction to the Study of Education** David Matheson, 2014-01-14 What is education This core

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