

Ideal Gas Law

Chapter 9

1. A gas at a pressure of 1 atm and a volume of 10 liters has a temperature of 25°C. What is the number of moles of the gas present?
2. A sample of gas has a volume of 5 liters, a pressure of 2 atm, and a temperature of 300 K. What is the number of moles of the gas present?
3. A gas has a pressure of 1.5 atm, a volume of 20 liters, and a temperature of 25°C. What is the number of moles of the gas present?
4. A gas at a volume of 30 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
5. A gas at a pressure of 1 atm, a volume of 25 liters, and a temperature of 0°C has a number of moles equal to 2. What is the pressure of the gas at a volume of 15 liters and a temperature of 27°C?
6. A gas at a pressure of 2 atm, a volume of 10 liters, and a temperature of 25°C has a number of moles equal to 4. What is the pressure of the gas at a volume of 20 liters and a temperature of 27°C?
7. A gas at a volume of 5 liters and a temperature of 27°C has a pressure of 1 atm. What is the number of moles of the gas present?
8. A gas at a pressure of 3 atm, a volume of 20 liters, and a temperature of 25°C has a number of moles equal to 5. What is the pressure of the gas at a volume of 10 liters and a temperature of 27°C?
9. A sample of gas has a volume of 15 liters, a pressure of 1.5 atm, and a temperature of 300 K. What is the number of moles of the gas present?
10. A gas at a volume of 10 liters and a temperature of 27°C has a pressure of 2 atm. What is the number of moles of the gas present?
11. Note: These questions are meant to be examples to guide you in creating your own worksheet. The specific answers to these questions may vary based on your textbook or course material.

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Terry C. Jones



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